

# 8.0% Efficient Sub-Micron $\text{CuIn}(\text{S},\text{Se})_2$ Solar Cells on $\text{Sn}:\text{In}_2\text{O}_3$ Back Contact via a Facile Solution Process

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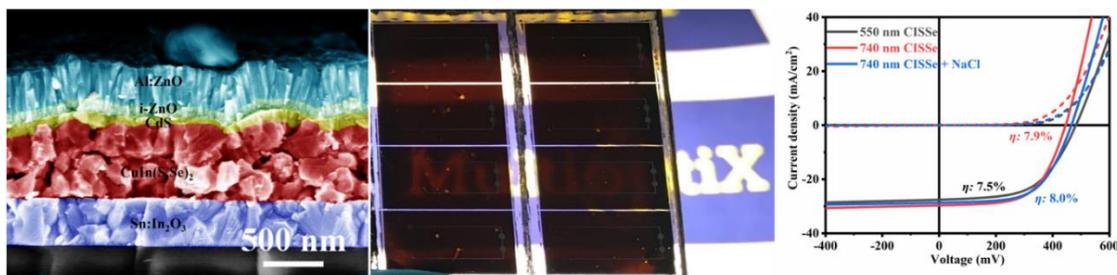
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## Graphical abstract



## Abstract

High-performance chalcopyrite solar cells have been fabricated on transparent conductive oxide (TCO) back contact through an environmentally benign solution, showing great potential in bifacial application. Ultra-thin (around 550 nm)  $\text{CuIn}(\text{S},\text{Se})_2$  (CISSe) solar cells were successfully deposited on  $\text{Sn}:\text{In}_2\text{O}_3$  (ITO) back contact via spin-coating of metal-chloride DMF (N, N-Dimethylformamide) solution, followed by selenization. The ultra-thin devices achieved a conversion efficiency of 7.5% when the absorber layer was selenized at 520 °C. With the increase in the absorber thickness to sub-micron (740 nm), the solar cells exhibited not only a higher short-circuit current density but also an improved fill factor compared to the ultra-thin devices, which resulted in an efficiency enhancement to 7.9%. Furthermore, NaCl solution pre-selenization treatment was demonstrated to improve the

performance of CISSe solar cells. When the sub-micron absorber was subject to 1 M NaCl solution prior to selenization, an 8.0% efficient CISSe device was achieved. To the best of our knowledge, this is the topmost performance for sub-micron CISSe solar cells fabricated from solution-based precursors on TCO back contact.

**Keywords:** CISSe solar cells, sub-micron absorber, ITO back contact, spin-coating, solution processing, NaCl pre-selenization treatment

## 1. Introduction

Photovoltaics is considered one of the most promising energy-harvesting technologies because it can directly convert the basically unlimited resource of solar energy to electrical energy. Copper indium gallium diselenide-based (CIGSe-based) solar cells with a champion efficiency of 23.35% were achieved with a 2-3  $\mu\text{m}$  thick absorber layer. [1] However, the thick CIGSe absorber relates to high cost e.g. due to the large consumption of scarce elements (In and Ga). Therefore, thinning the absorber layer would be conducive to reducing the absorber deposition duration and lessening the usage of scarce elements, simultaneously increasing production throughput and reducing manufacturing costs. [2-7]

Generally, molybdenum (Mo) is employed as a back contact for high-efficiency CIGSe solar cells because a  $\text{MoSe}_2$  interlayer between Mo back contact and CIGSe absorber will form during the absorber growth process and lead to a quasi-Ohmic contact. [8] However, due to the poor back reflectivity of the CIGSe/Mo interface, ultra-thin (around 500 nm) CIGSe solar cells suffer from dramatic optical losses. In addition, the opacity of the Mo back-contact limits their application in bifacial and semi-transparent structures. [9] Therefore, replacing opaque Mo with a transparent conductive oxide (TCO) as back contact is promising. In

previous reports, Al:ZnO (AZO), F:In<sub>2</sub>O<sub>3</sub> (FTO), and Sn:In<sub>2</sub>O<sub>3</sub> (ITO) have been used to replace the Mo back contact of CIGSe solar cells fabricated via a co-evaporation process. [5, 7, 10, 11] However, co-evaporation deposition is an expensive fabrication technique requiring a high vacuum environment. Therefore, the exploitation of a cost-effective and convenient process for CIGSe solar cell fabrication is crucial.

In past decades, the solution-based process has been proven a promising technique for chalcopyrite solar cell growth. The champion PCE (photovoltaic conversion efficiency) of Cu(In,Ga)(S,Se)<sub>2</sub> (CIGSSe) solar cells fabricated by a hydrazine-based solution process is 18.1%. [12] However, hydrazine is a highly toxic solution, which prohibits its industrial production for health, environmental and safety concerns. Jiang et al. have proven PCEs of 14.5 % for CISSe and 15.2 % for CIGSSe solar cells with 1.2 μm absorbers. These were obtained via a solution-based spin-coating process with metal chloride-DMF (DMF: N, N-Dimethylformamide) precursor ink, indicating that metal-chloride DMF as a precursor solution shows high potential in CIGSe-based solar cell fabrication. [13-15] A CIGSe device with an efficiency of 13.8% was achieved successfully via an all solution-processed fabrication (except for the Mo back contact). [16] On ITO back contact, Barange et al. reported 5.68% efficiency for CIGS via a spin-coating process of sol-gel solution. [17] Sousa et al. reported 6.1 % efficiency for CIGSe on FTO back contact via a screen-printing process. [18] However, the solar cell properties exhibited poor homogeneity.

Here, we first demonstrate 7.5% efficiency for ultra-thin CISSe solar cells (550 nm absorber) deposited on ITO back contact from metal-chloride DMF solution, which is the highest efficiency of ultra-thin CISSe solar cells fabricated on TCO back contact via a solution process. We find that the selenization temperature is a key for achieving high crystallinity,

morphology and homogeneity of the CISSe absorbers. To improve the efficiency of the CISSe solar cells, three different thicknesses (550 nm, 740 nm and 1440 nm CISSe) of the absorber layer are investigated. The absorbers with thicknesses of 550 nm, 740 nm, and 1400 nm are referred to as ultra-thin, sub-micron, and micron absorbers, respectively. A corresponding terminology is used for the related CISSe solar cells. The sub-micron CISSe solar cells show the best photovoltaic properties. The efficiency of the CISSe solar cells with a 740 nm absorber is further improved after being subject to a pre-selenization treatment with 1 M NaCl solution. 8.0% PCE of sub-micron CISSe solar cells with NaCl solution pre-selenization treatment is achieved, which is the topmost efficiency for sub-micron CISSe solar cells fabricated from solution-based precursors on TCO back contact. These high-quality sub-micron CISSe solar cells show great potential for bifacial applications.

## **2. Experimental**

### **2.1 ITO back contact deposition**

ITO back contacts of 400 nm thickness were deposited on SLG (soda-lime glass) substrate by DC-sputtering in a PRO Line PVD 75 (Kurt J. Lesker Company). It was reported that the electrical and optical performances of ITO back contact can be improved by annealing. [19] The ITO back contacts are subject to a heating treatment at 350 °C for 10 min in air before spin-coating precursor solution. The sheet resistances and optical properties are shown in Figure S1. Sheet resistances of ITO without and with annealing are quite comparable. Pre-annealing improves the transmission in the wavelength range from 300-600 nm and increases absorption from 500-1200 nm. The pre-annealing process can improve the wetting property between ITO and precursor solution, which is favourable for the formation of a flat film.

### **2.2 Fabrication of CISSe solar cells**

For the CuCl-InCl<sub>3</sub>-Thiourea (TU) solution, 1.197 g thiourea (TU, 99%, Alfa Aesar) was firstly added to 8 mL N, N-Dimethylformamide (DMF, 99.8%, Sigma-Aldrich) solvent to form a clear solution after 20 min stirring. Then, 0.253 g CuCl (99.8%, Sigma-Aldrich) was added to this solution and stirred until CuCl completely dissolved. Finally, 0.608 g InCl<sub>3</sub> (99.8%, Sigma-Aldrich) was added and a yellowish solution was obtained after overnight stirring. The solution was filtered by using a 0.45 μm polytetrafluoroethylene filter before spin-coating. To improve the wetting property of the ITO back contact with the precursor solution, the ITO back contacts were annealed at 350 °C for 10 min before spin-coating. The solution was deposited on ITO substrates via spin-coating at 1500 rpm for 60 s. Then the films were immediately annealed on a hot plate at 350 °C for 2 min and moved to a ceramic plate for cooling down naturally. The coating-annealing-cooling cycle was repeated 5 times, 8 times, and 15 times to make a total film thickness of around 550 nm (ultra-thin), 740 nm (sub-micron), and 1440 nm (micron), respectively. The as-prepared Cu-In-S precursor films were placed into a tube furnace with an Ar atmosphere and 250 mg Se powder. They were then subject to a first heating step at 350 °C for 15 min and a second step at an enhanced temperature for 20 min to form CISSe thin films. For the pre-selenization sodium treatment, the as-prepared sub-micron Cu-In-S precursor films were soaked in 0.4 M, 0.8 M or 1 M NaCl aqueous-ethanol solution (the volume ratio of water to ethanol is 1:1) for 10 min, respectively. [20] After selenization, the as-annealed films were etched in a 10% KCN solution for the removal of Cu<sub>2-x</sub>(S,Se). Then, an 80 nm CdS buffer layer was deposited onto the CISSe thin films by chemical bath deposition, and an 80 nm i-ZnO / 300 nm Al:ZnO bilayer was sputtered successively. A Ni/Al top grid was deposited by thermal evaporation. The active area of each CISSe solar cell was 0.5 cm<sup>2</sup> defined by mechanical scribing.

### **2.3 Characterization**

Grazing incidence X-ray diffraction (GIXRD) with an incidence angle of  $0.5^\circ$  was applied to measure the structure of the as-prepared films. The data was collected employing an X'Pert PRO diffractometer (PANalytical) with Ni-filtered Cu  $K\alpha$  radiation and a PIXcel detector. The morphology of the photovoltaic devices was measured by a LEO 1500 scanning electron microscope (SEM). The composition of the CISSe absorbers was extracted by a calibrated ED-XRF (Energy-dispersive X-ray fluorescence) of type SPECTRO XEPOS. The element profiles of the absorbers were identified by GDOES (glow discharge optical emission spectroscopy) on the instrument SPECTRO XEPOS. The current density-voltage ( $J$ - $V$ ) curves were measured under standard test conditions (AM1.5;  $100 \text{ mW/cm}^2$ ;  $25^\circ\text{C}$ ) by a WACOM sun-simulator containing both a Xenon and a halogen lamp. The external quantum efficiency (EQE) was measured on a home-built system applying calibrated Si and Ge diodes as references. In addition, the Capacitance-voltage ( $C$ - $V$ ) curves were taken under dark conditions with a BK PRECISION Model 895 operating at 100 kHz and a 5 mV testing signal.

### 3. Results

#### 3.1 Effect of selenization temperature

Figure 1a shows the GIXRD patterns of ultra-thin CISSe thin film solar cells. The positions of Bragg reflections at  $26.9^\circ$ ,  $44.7^\circ$  and  $53.1^\circ$  correspond to the (112)/(103), (204)/(220), and (116)/(312) planes of chalcopyrite, respectively. [13] Since the radius of the selenium atom is larger than that of sulphur, the positions of these Bragg reflections are right shifted compared to pure  $\text{CuInSe}_2$  and left-shifted compared to pure  $\text{CuInS}_2$ . Bragg reflections located at  $30.5^\circ$  and  $50.9^\circ$  are attributed to the ITO back contact. Additionally, the Bragg reflections pointing to the ZnO window layer are observed at  $34.4^\circ$ ,  $36.2^\circ$  and  $47.6^\circ$ . When the selenization is carried out at  $T = 500^\circ\text{C}$ , additional Bragg reflections of ITO at  $35.4^\circ$  and  $60.7^\circ$  appear.

Compared to  $T = 500\text{ }^{\circ}\text{C}$ , the XRD diffraction patterns for absorbers selenized at  $T \geq 520\text{ }^{\circ}\text{C}$  show a slight right shift (Figure 1b), especially at high diffraction angles where the shift of the Bragg reflections is more pronounced. This result indicates that as-prepared absorber layers at  $T \geq 520\text{ }^{\circ}\text{C}$  have a higher S/(S+Se) ratio, which is confirmed by XRF measurements as shown in Table S1. [21] However, for the CISSe absorber obtained at  $540\text{ }^{\circ}\text{C}$ , the shift of XRD patterns is not as obvious as for the absorbers at  $520\text{ }^{\circ}\text{C}$  and  $560\text{ }^{\circ}\text{C}$ . This observation is also in agreement with a lower S/(S+Se) ratio, see table S1.

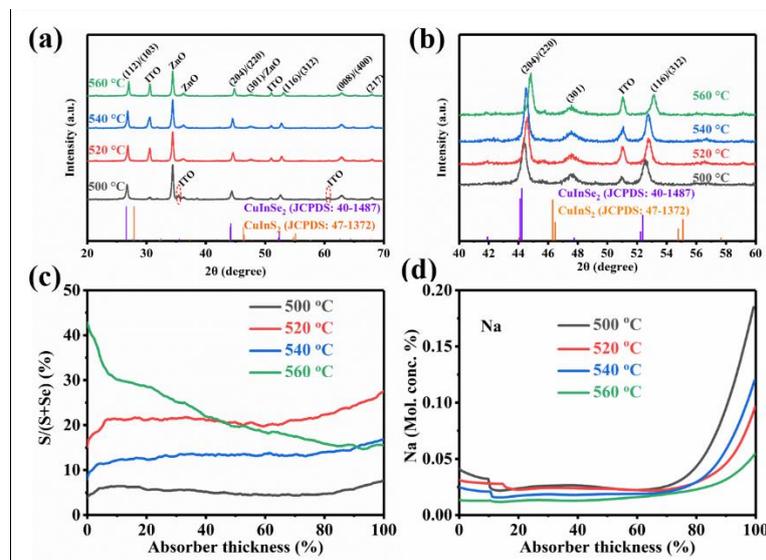


Figure 1 (a) and (b) XRD pattern of the CISSe solar cell devices selenized at different temperatures; (c) S/(S+Se) ratios and (d) Na content of the absorber layers measured by GDOES.

The S/(S+Se) depth profiles of the CISSe absorbers are measured via GDOES and are shown in Figure 1c. With increasing the selenization temperature above  $500\text{ }^{\circ}\text{C}$ , the content of S increases. It indicates that in the process of selenization at higher temperatures, the higher Se pressure suppresses the evaporation of S, accordingly inhibiting the replacement of Se by S. When  $T \leq 540\text{ }^{\circ}\text{C}$ , the S content at the CISSe back interface is higher than at the front interface. In addition, when  $T = 540\text{ }^{\circ}\text{C}$ , the content of S is lower than for  $T = 520\text{ }^{\circ}\text{C}$  and  $T =$

560 °C, which is in agreement with the position of the XRD peaks (Figure 1b). However, when increasing the temperature to  $T = 560$  °C, the opposite dependency is observed regarding the S/(S+Se) ratio. There is a factor of two higher S content at the front compared to the back interface. This observation may be explained by large grains in the CISSe layer forming rapidly under high Se pressure at high temperatures. However, S will volatilize along the grain boundary of this large-grain layer, leading to inhomogeneous grains at the front interface of the CISSe absorber (Figure 2d). The near front surface consists of large CISSe grains and an aggregation of small CISSe grains. Large CISSe grains form at the rear interface of the CISSe absorber. In general, the grain size of CISSe with a Se-rich composition is larger than the CISSe with S-rich. [8] Therefore, a higher S/(S+Se) ratio is observed at the front surface decreasing towards the rear interface. The GDOES results are in good agreement with XRD and XRF measurements (Table S1). Na exhibits a similar trend in these solar cells (Figure 1d). A high Na content exists near the back interface, then it decreases towards the front surface. A high selenization temperature can promote the diffusion of Na from the substrate to the absorber layer. However, with increasing selenization temperature, the Na content of the absorber decreases in our experiments (Figure 1d). The reason may be a removal of Na near the surface during the KCN etching and the CdS deposition process. Na can stay at the grain boundary and grain interior. [22] The higher Na content for 500 °C selenization temperature is attributed to its significantly smaller grains and higher density of grain boundaries. [13]

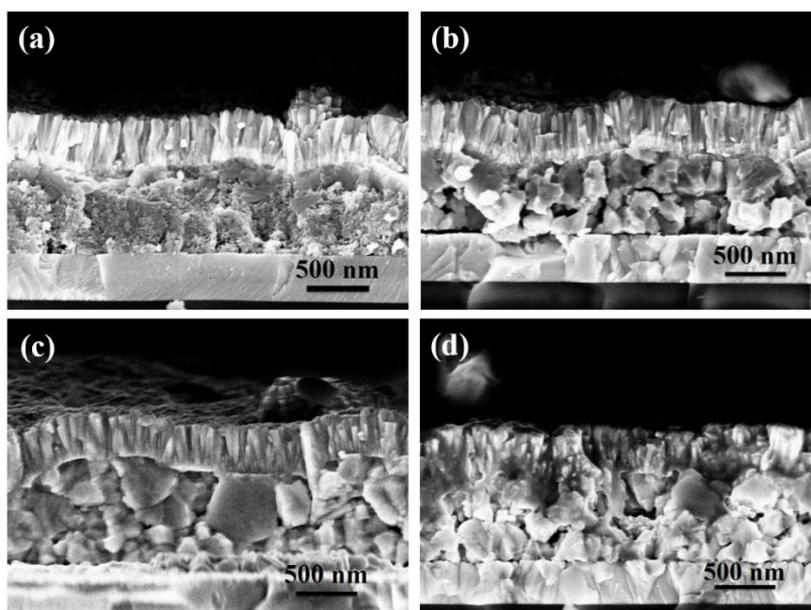


Figure 2 SEM images of CISSe solar cells prepared at different selenization temperatures: (a) 500 °C, (b) 520 °C, (c) 540 °C, and (d) 560 °C.

The cross-section scanning electron microscopy (SEM) images of the CISSe devices with varied selenization temperatures are shown in Figure 2. For  $T = 500$  °C, the CISSe absorber shows a double-layer structure with a thick sub-layer of fine grains at the bottom and a very thin sub-layer with large grains on top, implying that the low selenization temperature is disadvantageous to CISSe grain growth. When the selenization temperature increases to  $T = 520$  °C, larger grains of around 300 - 500 nm size are observed. However, voids still exist in the absorber layer and at the back contact interface. The largest CISSe grains with diameters of around 500 - 650 nm form when the precursor film selenization is carried out at 540 °C. In this case, the large grains of the absorber layer are densely packed, almost absent of voids. When the selenization temperature increases to 560 °C, a CISSe absorber layer with a triple-layer structure is observed: large grains at the bottom and on top, and fine grains in the middle. However, the top layer consists of large CISSe grains and an aggregation of small CISSe grains (Figure 2d, left: fine grains aggregation; right: large grains). This

inhomogeneous morphology can be ascribed to the difference in element distribution. A large amount of liquid Se penetrates the whole film under high Se vapour pressure, and Se bonds with Cu to form  $\text{Cu}_{2-x}\text{Se}$ , which facilitates the formation of large grains and denser films. [13,23] However, the S tends to stronger volatilize outflow when the precursor films are selenized at 560 °C. Additionally, the as-formed dense large grains will inhibit the S outflow (Figure 2d). [23] Sulfur can only evaporate in the area containing abundant grain boundaries, leading to the formation of S-rich CISSe grains with small grain size after selenization. Even though large grains are located at both the bottom and top of the absorber, the grain size of the bottom layer is more homogeneous than that of the top one. A possible reason is that bottom CISSe grains have a high content of Se and a more homogeneous elemental composition than CISSe grains on top of the absorber (Figure 1c).

Figure 3a shows the  $J$ - $V$  curves of the best CISSe solar cells under AM 1.5G illumination, and the corresponding photovoltaic parameters are summarized in Table 1. When the selenization is carried out at  $T = 500$  °C, a maximum power conversion efficiency (PCE) of 4.8 % is achieved with a short-circuit current density ( $j_{sc}$ ) of 24.8 mA/cm<sup>2</sup>, an open-circuit voltage ( $V_{oc}$ ) of 384.1 mV, and a fill factor ( $FF$ ) of 50.3%. The PCE at  $T = 520$  °C is remarkably enhanced to 7.5%. The efficiency improvement can be explained by the better crystallinity at high temperatures. However, when the selenization temperature increases to  $T = 540$  °C and  $T = 560$  °C, the PCEs decrease to 6.6%, and 5.9%, respectively. The same trend can be observed in  $j_{sc}$  and  $FF$ , the highest  $j_{sc}$  of 27.6 mA/cm<sup>2</sup> and  $FF$  of 56.1% are obtained at 520 °C. However, the highest  $V_{oc}$  value of 492.2 mV achieves when the temperature increases to 560 °C. The higher  $V_{oc}$  can be attributed to the better crystallinity and the higher content of S observed at higher selenization temperatures. Simultaneously, due to the wider band gap of the CISSe absorber with higher S content,  $j_{sc}$  is lowered. [8, 12] The series resistance ( $R_s$ ) values of the samples are calculated from the plots of  $dV/dJ$  versus  $1/(J+J_{sc})^2$  (shown in

Figure S3). With increasing selenization temperature  $R_s$  increases from  $1.7 \Omega \text{ cm}^2$  (for  $500 \text{ }^\circ\text{C}$ ) to  $2.2 \Omega \text{ cm}^2$  (for  $520 \text{ }^\circ\text{C}$ ),  $4.3 \Omega \text{ cm}^2$  (for  $540 \text{ }^\circ\text{C}$ ), and  $5.8 \Omega \text{ cm}^2$  (for  $560 \text{ }^\circ\text{C}$ ), respectively. The enhancement of  $R_s$  with increasing absorber selenization temperature ascribes to the formation of a thin  $\text{In}_2\text{O}_3$  layer at the CISSe/ITO interface during the high-temperature selenization owing to the low standard molar enthalpy of  $\text{In}_2\text{O}_3$ . [24]

As shown in figure 3b, the  $V_{oc}$  of the solar cells increases with increasing selenization temperature. However, when the precursor films are selenized at  $T = 540 \text{ }^\circ\text{C}$ , the  $V_{oc}$  of the corresponding solar cells slightly decreases. This variation tendency of  $V_{oc}$  can be related to the content of S in the CISSe absorbers (Table S1). Increasing the selenization temperature results in the decrease of  $j_{sc}$  (Figure 3c) because the CISSe absorber has a wider band gap in high selenization temperature due to the increase of S content (Table S1). Both  $FF$  and PCE increase with increasing selenization temperature to  $T = 520 \text{ }^\circ\text{C}$  and then decrease with further increasing selenization temperature (Figure 3d and 3e). These results indicate that  $T = 520 \text{ }^\circ\text{C}$  is favourable for fabricating high-quality CISSe absorbers on ITO back contact for semi-transparent CISSe solar cell applications.

The external quantum efficiency (EQE) spectra of the best devices are shown in Figure 3f and the optical properties are shown in Figure S2. These EQEs are higher than 65% in the visible wavelength range and the intensity of the EQE response surpasses 75%. However, the EQE results are relatively low because of the thin absorber layer (550 nm). The device applying the absorber selenized at  $T = 560 \text{ }^\circ\text{C}$  shows a higher EQE in the range from 380-950 nm wavelength compared to the samples of other annealing temperatures. However, the EQE drops rapidly when the wavelength is longer than 1000 nm, owing to a higher content of S in the absorber selenized at  $T = 560 \text{ }^\circ\text{C}$  compared to those selenized at  $T < 560 \text{ }^\circ\text{C}$  (Table S1). The EQE spectra of all devices are overlapping in the range of 300 - 380 nm wavelength owing to the absorption originating from the CdS/i-ZnO/AZO layers. [7]

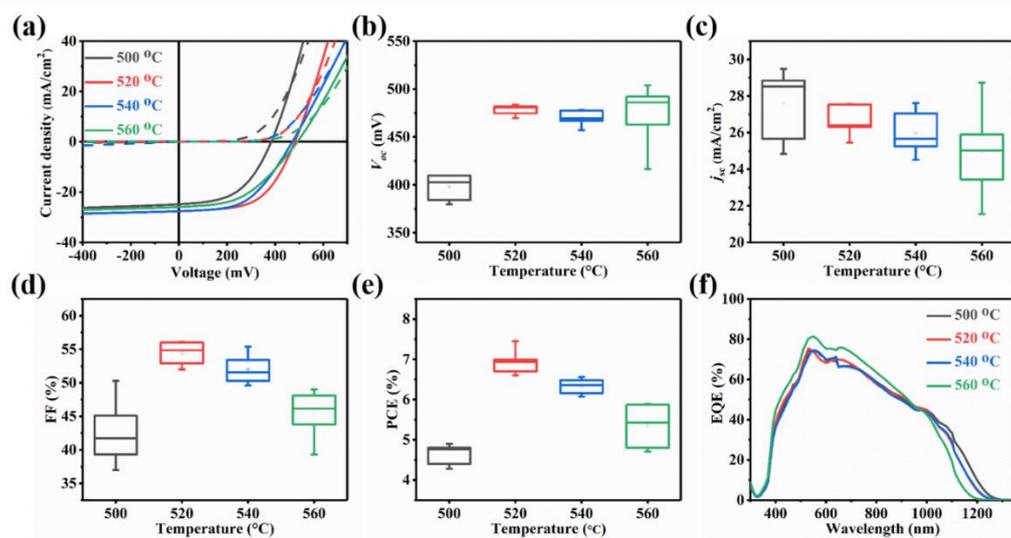


Figure 3 (a)  $J$ - $V$  curves of the best CISSe devices grown on ITO back contact at various selenization temperatures. The distribution (each sample contained 6 cells) of (b) open circuit voltage ( $V_{oc}$ ), (c) short circuit current ( $j_{sc}$ ), (d) fill factor ( $FF$ ), and (e) maximum power conversion efficiency (PCE). (f) External quantum efficiency (EQE) of the best CISSe devices.

Table 1 Photovoltaic parameters averaged over 6 CISSe devices and for the highest efficiency CISSe device grown on ITO back contact at various selenization temperatures (corresponding to Figure 3).

Selenization Temperature	$V_{oc}$ (mV)	$j_{sc}$ (mA/cm <sup>2</sup> )	$FF$ (%)	PCE (%)	$R_s$ ( $\Omega$ cm <sup>2</sup> )
500 °C	398.1±13.1	27.6±1.9	42.5±4.7	4.7±0.2	1.7
	384.1	24.8	50.3	4.8	
520 °C	478.6±5.3	26.6±0.8	54.5±1.7	6.9±0.3	2.2
	481.7	27.6	56.1	7.5	
540 °C	469.8±7.9	26.0±1.2	52.0±2.2	6.3±0.2	4.3
	471.7	27.6	50.3	6.6	
560 °C	474.6±31.6	24.9±2.5	45.4±3.5	5.4±0.6	5.8
	492.2	25.9	46.1	5.9	

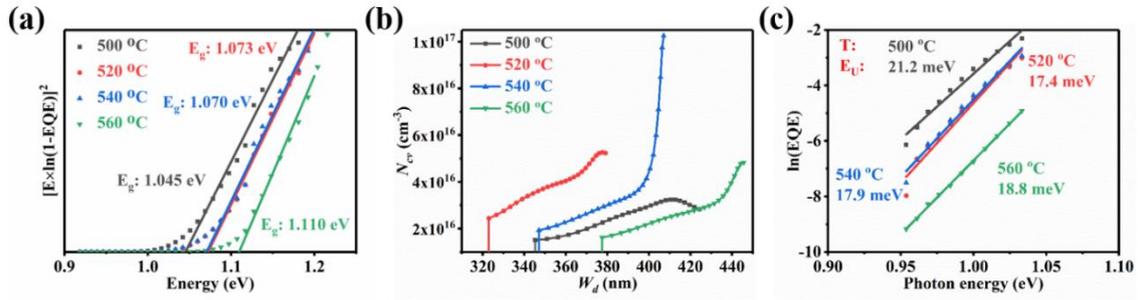


Figure 4 (a) Band gaps of devices with different selenization temperatures calculated from EQE data. (b) charge carrier density and depletion width derived from  $C$ - $V$  curves. (c)  $\ln(\text{EQE})$  vs photon energy at the long-wavelength edge to determine the  $E_U$  values for CISSe devices with different selenization temperatures.

From the EQE spectra of the devices with the highest efficiency, band gaps can be calculated (see Figure 4a). The band gaps of these CISSe devices extracted from the  $[E \times \ln(1 - \text{EQE})]^2$  vs energy plots are shown in Figure 4a. With increasing selenization temperature, wider band gaps are observed. When  $T = 500$  °C, a 1.04 eV band gap is obtained. With increasing the selenization temperature to  $T = 520$  °C and  $T = 540$  °C, similar band gap values (1.07 eV for 520 °C and 1.08 eV for 540 °C) are observed from EQE calculation results. The absorber carrying out selenization at  $T = 560$  °C shows the widest band gap of 1.11 eV. These results are in good agreement with the  $V_{oc}$  improvement at higher selenization temperatures as compared to  $T = 500$  °C.

The profiles of carrier concentration ( $N_{cv}$ ) vs depletion width ( $W_d$ ) are calculated from  $C$ - $V$  curves (Figure 4b). [7]  $N_{cv}$  of  $1.5 \times 10^{16}$  and  $W_d$  of 345.3 nm are obtained when the CISSe absorber is selenized at 500 °C. When the selenization temperature increases to 520 °C,  $N_{cv}$  increases to  $2.3 \times 10^{16}$  cm<sup>-3</sup> and  $W_d$  thus decreases to 323.1 nm, respectively. However, with further increasing selenization temperature,  $N_{cv}$  decreases from  $2.3 \times 10^{16}$  cm<sup>-3</sup> ( $T = 520$  °C) to

$1.9 \times 10^{16} \text{ cm}^{-3}$  ( $T = 540 \text{ }^\circ\text{C}$ ) and  $1.6 \times 10^{16} \text{ cm}^{-3}$  ( $T = 560 \text{ }^\circ\text{C}$ ). Correspondingly,  $W_d$  increases from 323.1 nm ( $T = 520 \text{ }^\circ\text{C}$ ) to 347.1 nm ( $T = 540 \text{ }^\circ\text{C}$ ) and 377.5 nm ( $T = 560 \text{ }^\circ\text{C}$ ), respectively. It is reported that the high-quality CIGSe solar cells have a high carrier concentration and a related narrow depletion width. [25, 26] As a result, the highest values of performance parameters are obtained for devices selenized at 520 °C.

Figure 4c shows the relation of  $\ln(\text{EQE})$  vs photon energy at the long wavelength edge (1200 - 1300 nm). From this, the Urbach energy ( $E_U$ ) can be extracted, which affects the carrier mobility and lifetime of the solar cells. [27, 28]  $E_U$  decreases from 20.2 meV ( $T = 500 \text{ }^\circ\text{C}$ ) to 17.4 meV ( $T = 520 \text{ }^\circ\text{C}$ ) with a related increase in PCE from 4.8% to 7.5%, because the enlargement of CISSe grain size is obtained at  $T = 520 \text{ }^\circ\text{C}$ , leading to good absorber quality. [28] However, when the selenization temperature continues to increase,  $E_U$  increases to 17.9 meV ( $T = 540 \text{ }^\circ\text{C}$ ,  $\eta = 6.6\%$ ) and 18.8 meV ( $T = 560 \text{ }^\circ\text{C}$ ,  $\eta = 5.9\%$ ), respectively. It indicates that an even higher temperature is not beneficial for preparing high-efficiency CISSe solar cells on ITO back contact. Conclusively, the best solar cell properties are achieved for selenization at  $T = 520 \text{ }^\circ\text{C}$ . Therefore, this temperature is used for further research in this work.

### 3.2 Effect of absorber thickness and NaCl treatment

The absorber thickness can strongly affect the solar cell properties. [9] Thinner absorber layers result in low cost because they reduce material consumption and increase production throughput. However, the efficiency of thin absorbers is limited by electrical and optical losses, such as increased recombination at back contact and reduced light absorption. [3-6, 29] As shown in the previous section, the best solar cell properties are obtained for absorbers selenized at 520 °C, thus  $T = 520 \text{ }^\circ\text{C}$  will be used for exploring the effect of absorber thickness. Three absorbers thicknesses are utilized for the investigations: 550 nm (ultra-thin absorber), 740 nm (sub-micron absorber), and 1440 nm (micron absorber). The best device

with a sub-micron CISSe absorber exhibits an efficiency of 7.9%, which is 0.4% (abs.) higher than the ultra-thin CISSe device (Figure 5a). However, ultra-thin CISSe shows a higher  $V_{oc}$  of 481.7 mV compared to 445.6 mV for the sub-micron CISSe, whereas the opposite trend is observed for  $j_{sc}$  and  $FF$ . Previous reports have proven that the fabrication of thick absorbers via a solution process will lead to carbon residuals and voids at the interface between the absorber and the back contact. [8, 30] These defects located at the rear interface will increase the recombination and result in a lower  $V_{oc}$ . The increase in  $j_{sc}$  of sub-micron solar cells can be attributed to higher light absorption due to the thicker absorber. CISSe solar cells with a micron absorber show inferior properties and poor uniformity (Figure S4), the highest PCE of these devices is 1.8%.  $V_{oc}$ ,  $j_{sc}$ , and  $FF$  are significantly reduced to 376.0 mV, 16.4 mA/cm<sup>2</sup> and 29.8%, respectively. The absorbers easily peel off from the ITO back contact in this case. The series resistance decreases from 2.2  $\Omega$  cm<sup>2</sup> for the ultra-thin absorber to 1.1  $\Omega$  cm<sup>2</sup> for the sub-micron absorber and then increases to 8.1  $\Omega$  cm<sup>2</sup> for the micron absorber (Figure S5 and Table 2). As the adhesion to the ITO back contact may become a challenge for thick absorbers and the best efficiency of CISSe solar cells on ITO was obtained for sub-micron absorbers, these are utilized for further research.

Indeed, compared to ultra-thin absorbers, increasing the absorber thickness to sub-micron (740 nm) can improve the CISSe solar cell properties. However, a lower  $V_{oc}$  is obtained for the thicker absorbers. It is known that Na incorporation can significantly increase the  $V_{oc}$  of CIGSe-based solar cells. [20, 26] The  $J$ - $V$  results of sub-micron CISSe suffering from a pre-selenization treatment with various concentrations of NaCl solution are shown in Figure S6 and Table S3. The corresponding electrical properties are shown in Figure S7. The efficiency increases to 8.0% using 1 M NaCl for pre-selenization treatment (Figure 5a). Additionally, the  $V_{oc}$  increases to the value of 466.0 mV, however, the  $j_{sc}$  and  $FF$  slightly decrease to 28.7 mA/cm<sup>2</sup>, and 59.6%, respectively. For the respective best performing devices, the reference

CISSe solar cell presents a higher  $j_{sc}$  and  $FF$  than the CISSe solar cells subject to Na treatment at various concentrations (Figure S6). Furthermore, the  $j_{sc}$  and  $FF$  of the best performing CISSe solar cells incorporating Na decrease with increasing the Na concentration before starting to recover (for 1 M NaCl).

The reference ultra-thin CISSe solar cell presents an EQE peaking at 74% for 540 nm wavelength. For the sub-micron CISSe solar cell, the EQE also peaks at 540 nm with a value surpassing 87%. The highest EQE is observed in the range from 380-1300 nm wavelength for sub-micron CISSe solar cells. The sub-micron CISSe solar cell with 1 M NaCl treatment has a higher EQE photo-response. A narrower band gap of the sub-micron CISSe solar cells (both with or without NaCl treatment) can be calculated from the EQE spectrum according to the formula  $[E \times \ln(1 - EQE)]^2$ .

Compared to the ultra-thin CISSe device, both the  $N_{cv}$  and  $W_d$  of sub-micron CISSe decrease. The sub-micron CISSe without NaCl treatment has a charge carrier density of  $1.9 \times 10^{16} \text{ cm}^{-3}$  and a  $W_d$  of 315.0 nm. The absorber with 1 M NaCl treatment exhibits a lower hole density ( $9.1 \times 10^{15} \text{ cm}^{-3}$ ) and the  $W_d$  decreases to 303.5 nm. The remarkable lower hole density should be attributed to higher recombination at the rear interface, reducing of  $V_{oc}$  and PCE. [7]

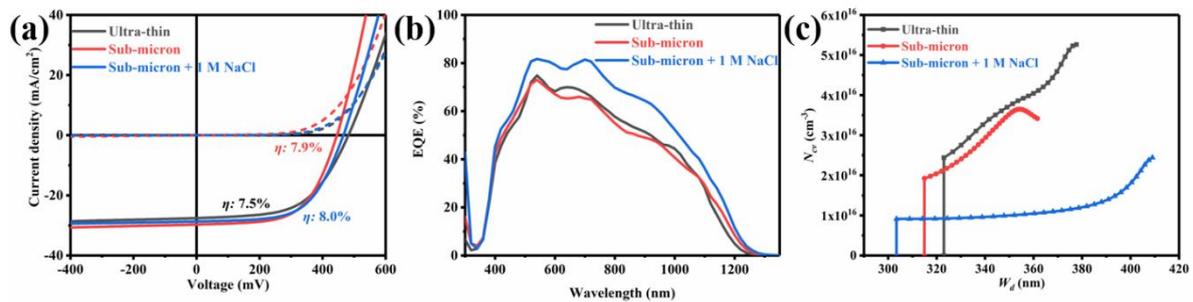


Figure 5 (a)  $J$ - $V$ , (b) EQE curves and (c)  $N_{cv}$ - $W_d$  relation of the best CISSe devices with different absorber thicknesses and with NaCl treatment grown on ITO back contact.

Table 2 Averaged (over six devices) and best CISSe photovoltaic device parameters for ultra-thin CISSe, Sub-micron CISSe and Sub-micron CISSe implementing 1 M NaCl solution treatment.

	$V_{oc}$ (mV)	$j_{sc}$ (mA/cm <sup>2</sup> )	$FF$ (%)	PCE (%)	$R_s$ ( $\Omega$ cm <sup>2</sup> )
Ultra-thin CISSe	478.6 $\pm$ 5.3	26.6 $\pm$ 0.8	54.5 $\pm$ 1.7	6.9 $\pm$ 0.3	2.2
	481.7	27.6	56.1	7.5	
Sub-micron CISSe	448.5 $\pm$ 2.6	29.1 $\pm$ 0.6	56.3 $\pm$ 3.2	7.3 $\pm$ 0.4	1.1
	445.6	29.7	59.8	7.9	
Sub-micron CISSe + 1 M NaCl	463.9 $\pm$ 5.9	28.2 $\pm$ 0.6	58.0 $\pm$ 2.0	7.6 $\pm$ 0.3	1.7
	466.0	28.7	59.6	8.0	

## 4. Discussion

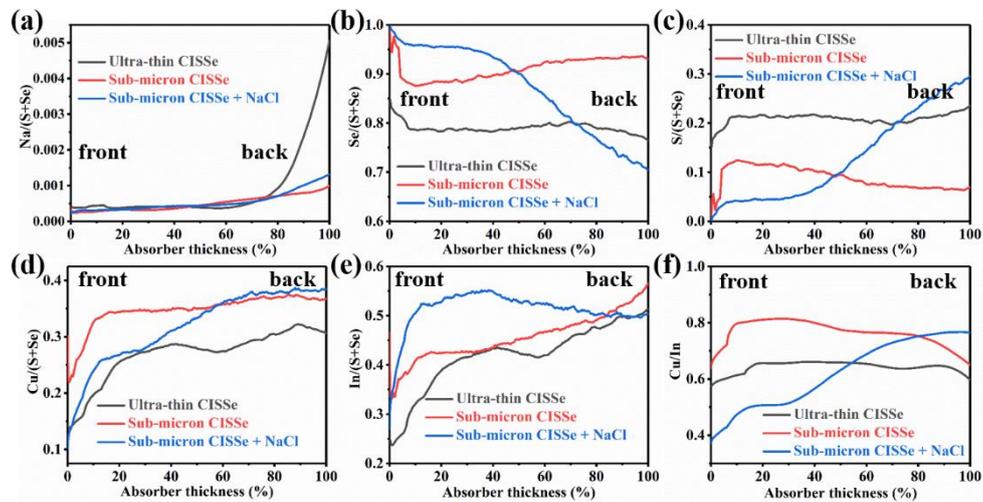


Figure 6 Composition details of ultra-thin (550 nm) and sub-micron (740 nm) CISSe absorber films without and with NaCl treatment measured by GDOES: (a) Na/(S+Se), (b) Se/(S+Se), (c) S/(S+Se), (d) Cu/(S+Se), (e) In/(S+Se), and (f) Cu/In ratios. The absorber thicknesses are normalized to 100% (corresponding to the absorber). The selenization was carried out at  $T = 520$  °C.

#### 4.1 Effect of absorber thickness

Ultra-thin (550 nm) and sub-micron (740 nm) CISSe absorbers have a similar content of Cu and In per unit thickness because these elements originate from the precursor solution and they will not volatilize seriously during the selenization process. Elemental S will volatilize during the selenization and be replaced by Se. All precursor films are subject to a selenization process utilizing the same content of Se. Therefore, the content of Se in various absorber thicknesses is different per unit thickness. In other words, the thin absorber has a higher Se/(total elements) ratio.

A higher Na/(S+Se) ratio exists near the rear interface of the ultra-thin absorber as compared to the sub-micron absorber (Figure 6a). The possible reason is that whilst the same amount of Na diffuses into the absorber from the glass substrate during the high-temperature selenization process, the thinner absorber obtains a higher Na content per unit thickness. The sub-micron absorber has a higher Se/(S+Se), Cu/(S+Se) and In/(S+Se) ratio compared to the ultra-thin absorber (Figure 6b, 6d, and 6e). These results can be explained by the formation of a dense large-grained layer at the front surface of the ultra-thin absorber. [13] It is reported that Na can promote the transformation of fine grains into large ones during selenization. The ultra-thin absorber has a shorter diffusion distance of Na than the sub-micron absorber.

Therefore, it is expected that the formation of a dense layer with large grains is facilitated in the ultra-thin absorber. This dense layer will inhibit the diffusion of Se towards the absorber and limit the volatilization of S. [13, 31] Thus, the result is that the ultra-thin absorber has a larger Se/(Se+S) than that of the sub-micron absorber. A higher Se/(S+Se) ratio is still observed at the front surface owing to the large-grained layer at near-surface (Figure 6b). [13] Correspondingly, the S/(S+Se) content is lower at the front surface and increases towards the back. As stated before, both ultra-thin and sub-micron precursor films are selenized in the same content of Se, so the ultra-thin absorbers have an overall higher content of Se/(S+Se).

As Cu and In are introduced through the precursor solution and, therefore, have similar concentrations, the ultra-thin absorbers have an overall lower Cu/(S+Se) and In/(S+Se) ratio than sub-micron absorbers. Both absorber types present a consistent tendency of Cu/In with higher Cu/In ratios in the bulk and lower at the front and rear interface. This increase in Cu/In ratios towards the bulk can be explained by forming a Cu-poor layer at the front interface (potential combination of  $\text{CuIn}(\text{S},\text{Se})_2$  and  $\text{CuIn}_3(\text{S},\text{Se})_5$ ). The decrease of Cu/In ratios at the rear interface can be ascribed to a small content of In loss owing to the formation of  $\text{In}_2(\text{S}_x,\text{Se}_{1-x})_3$  during the selenization. [24]

#### 4.2 Effect of NaCl treatment

Generally, the ultra-thin and sub-micron absorbers have a similar Na/(S+Se) ratio at the front surface because the excess Na is removed by a KCN etching process (Figure 6a). [32, 33] However, NaCl pre-selenization treatment will result in a redistribution of elements (Figure S8). Compared to the absorbers without NaCl treatment, the addition of Na results in a smaller value of Cu/(S+Se) (Figure 6d) and a larger value of In/(S+Se) (Figure 6e), leading to a smaller Cu/In ratio (Figure 6f). The Cu/In ratio near the surface of the sub-micron CISSe absorbers with a NaCl pre-selenization treatment is 0.37 only (Figure 6f), which can be explained by forming the copper-deficient ordered vacancy compound  $\text{CuIn}_3\text{Se}_5$  (the ratio of Se/(Se+S) is close to 100% at near-surface in Figure 6b). [32-34] This Cu-poor compound between the absorber and buffer layer can improve the CISSe solar cell performance, including a higher efficiency. When the absorber is subject to NaCl pre-selenization treatment, the Na can bond with Se to form  $\text{Na}_2\text{Se}_x$  which is beneficial for the formation of a dense layer of large grains at the absorber surface, inhibiting the diffusion of elemental Se towards the absorber and limiting the volatilization of S. [8, 13, 32, 33] Therefore, the absorber with NaCl pre-selenization treatment has a high Se/(S+Se) at the very front surface of the absorber and an overall inverse composition profile of S/(S+Se) (Figure 6b-c).

## 5. Conclusion

Ultra-thin and sub-micron CISSe solar cells are successfully fabricated on ITO back contact via a low-cost metal-chloride DMF solution process. Variations of selenization temperature are investigated. Increasing the selenization temperature above 520 °C may improve the crystallinity, morphology, and homogeneity of the CISSe absorbers. We demonstrate that 7.5 % efficiency for semi-transparent ultra-thin CISSe solar cells (550 nm CISSe absorber thickness) is obtained at 520 °C of selenization. So far, to the best of our knowledge, our results have achieved the highest performance for ultra-thin CISSe solar cells grown on ITO back contact via a solution process (Table S4). Furthermore, CISSe solar cells with different absorber thicknesses are studied. The efficiency of these solar cells firstly increases with the absorber thickness increasing to sub-micron (740 nm); however, it decreases when reaching micron (1440 nm) thickness. In addition, the CISSe solar cell can achieve an efficiency of 8% when the sub-micron absorber is subject to 1 M NaCl solution treatment. This work offers a facile method to prepare high-quality Na-doped CISSe films for integration in semi-transparent solar cell devices.

## Associated Content

### Supporting Information

Sheet resistance and optical properties of annealed ITO (Figure S1); XRF results (Table 2); optical properties (Figure S2) and series resistances (Figure S3) of ultra-thin CISSe absorbers; *J-V* curve of micron CISSe device (Figure S4); series resistance (Figure S5); PV parameters of NaCl treatment CISSe (Figure S6 and Table S2); fitting electrical parameters

(Figure S7 and Table S3); S/(S+Se) ratios of NaCl treatment CISSe absorbers (Figure 8); and overview on high performance TCO/CIGSSe-based solar cells (Table S4)

## Notes

The authors declare no competing financial interest.

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