

L1

add the following stock solutions to 900 ml of filtered seawater (0,1µm) and fill up to 1 liter

Components	Stock Solution	Addition per Litre of Culture Medium
1. NaNO_3	75.00 g / l dH ₂ O	1 ml
2. $\text{NaH}_2\text{PO}_4 \times \text{H}_2\text{O}$	5.00 g / l dH ₂ O	1 ml
3. Vitamin Solution		1 ml
Vitamin B12	0.20 mg / l dH ₂ O	
Biotin	1.00 mg / l dH ₂ O	
Thiamine-HCl	100.00 mg / l dH ₂ O	
Niacinamide	0.10 mg / l dH ₂ O	

pH of the Vitamin Solution should be around pH 7.0

4. Trace Metals 1 ml

4.1. Preparation of Trace Metal Solution

$\text{Na}_2\text{EDTA} \times 2 \text{H}_2\text{O}$: 4.36 g

$\text{FeCl}_3 \times 6 \text{H}_2\text{O}$: 3.15 g

Dissolve in 1000 ml dH₂O, then add 1 ml of Primary Trace Metals each (see below).

Primary Trace Metals are stored frozen as 1 ml aliquots.

4.2. Primary Trace Metals

4.2.1. K_2CrO_4	0.194 g / 100 ml dH ₂ O
4.2.2. $\text{CoCl}_2 \times 6 \text{H}_2\text{O}$	1.00 g / 100 ml dH ₂ O
4.2.3. $\text{CuSO}_4 \times 5 \text{H}_2\text{O}$	0.25 g / 100 ml dH ₂ O
4.2.4. $\text{MnCl}_2 \times 4 \text{H}_2\text{O}$	18.00 g / 100 ml dH ₂ O
4.2.5. $\text{Na}_2\text{MoO}_4 \times 2 \text{H}_2\text{O}$	1.89 g / 100 ml dH ₂ O
4.2.6. $\text{NiSO}_4 \times 6 \text{H}_2\text{O}$	0.27 g / 100 ml dH ₂ O
4.2.7. H_2SeO_3	0.13 g / 100 ml dH ₂ O
4.2.8. Na_3VO_4	0.184 g / 100 ml dH ₂ O
4.2.9. $\text{ZnSO}_4 \times 7 \text{H}_2\text{O}$	2.20 g / 100 ml dH ₂ O