



Seminar III

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- ATP-Generation via substrate level phosphorylation, electron transport phosphorylation
- CO₂ Fixation
- (Eukaryotic/prokaryotic cell)
- Bacterial cell wall
- Antibiotic resistance

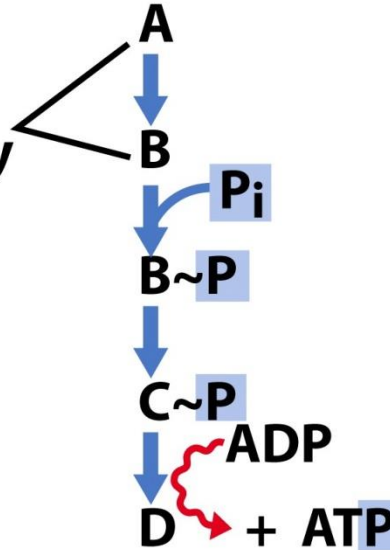
Basic Mechanisms of Energy Conservation

Substrate-level phosphorylation

Formation of energy-rich intermediates produces ATP

Intermediates in the biochemical pathway

Compound	G° kJ/mol
High energy	
Phosphoenolpyruvate	-51.6
1,3-Bisphosphoglycerate	-52.0
Acetyl phosphate	-44.8
ATP	-31.8
ADP	-31.8
Low energy	
AMP	-14.2
Glucose 6-phosphate	-13.8



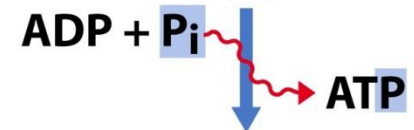
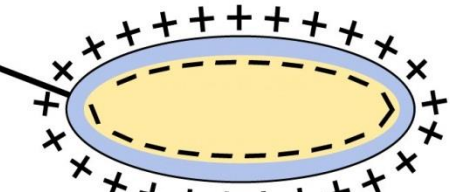
Substrate-level phosphorylation

Figure 5-13a Brock Biology of Microorganisms 11/e
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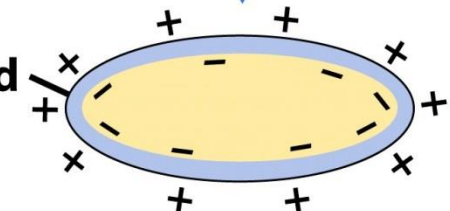
Elektron-transport phosphorylation

(Oxidative Phosphorylation)

Energized membrane



Less energized membrane



Oxidative phosphorylation

Figure 5-13b Brock Biology of Microorganisms 11/e
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Fig. 5.13 Brock Biology of Microorganisms (11th edition) (Madigan et al.)

Energy-Rich Compounds

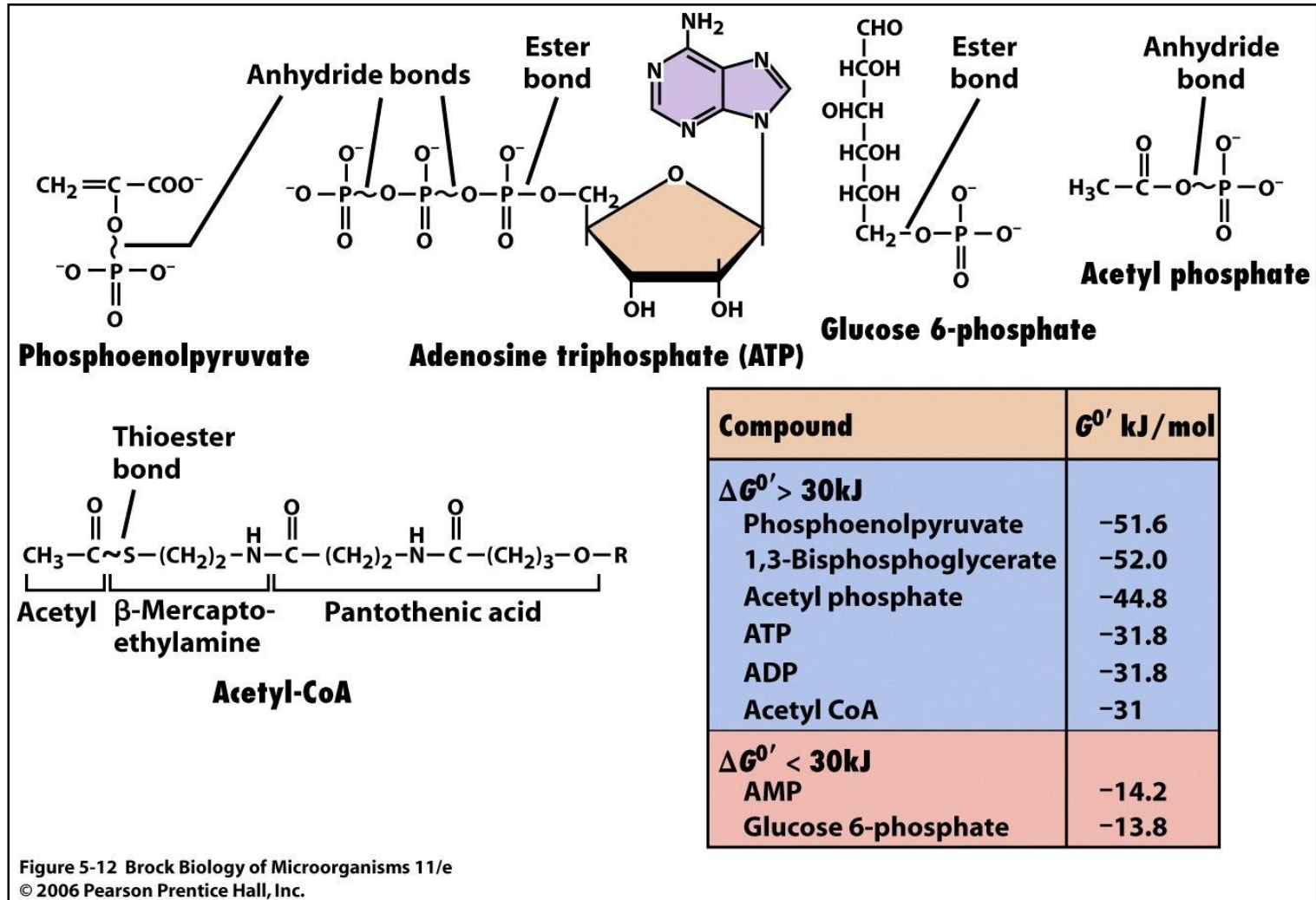


Figure 5-12 Brock Biology of Microorganisms 11/e
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Chemiosmotic Model

- In this simple representation of the chemiosmotic theory applied to mitochondria, electrons from NADH and other oxidizable substrates pass through a chain of carriers arranged asymmetrically in the inner membrane.
- Electron flow is accompanied by proton transfer across the membrane, producing both a chemical gradient (ΔpH) and an electrical gradient ($\Delta\psi$).
- The inner mitochondrial membrane is impermeable to protons; protons can reenter the matrix only through proton-specific channels (F_o). The **proton-motive force (PMF)** that drives protons back into the matrix provides the energy for ATP synthesis, catalyzed by the F_1 complex associated with F_o .

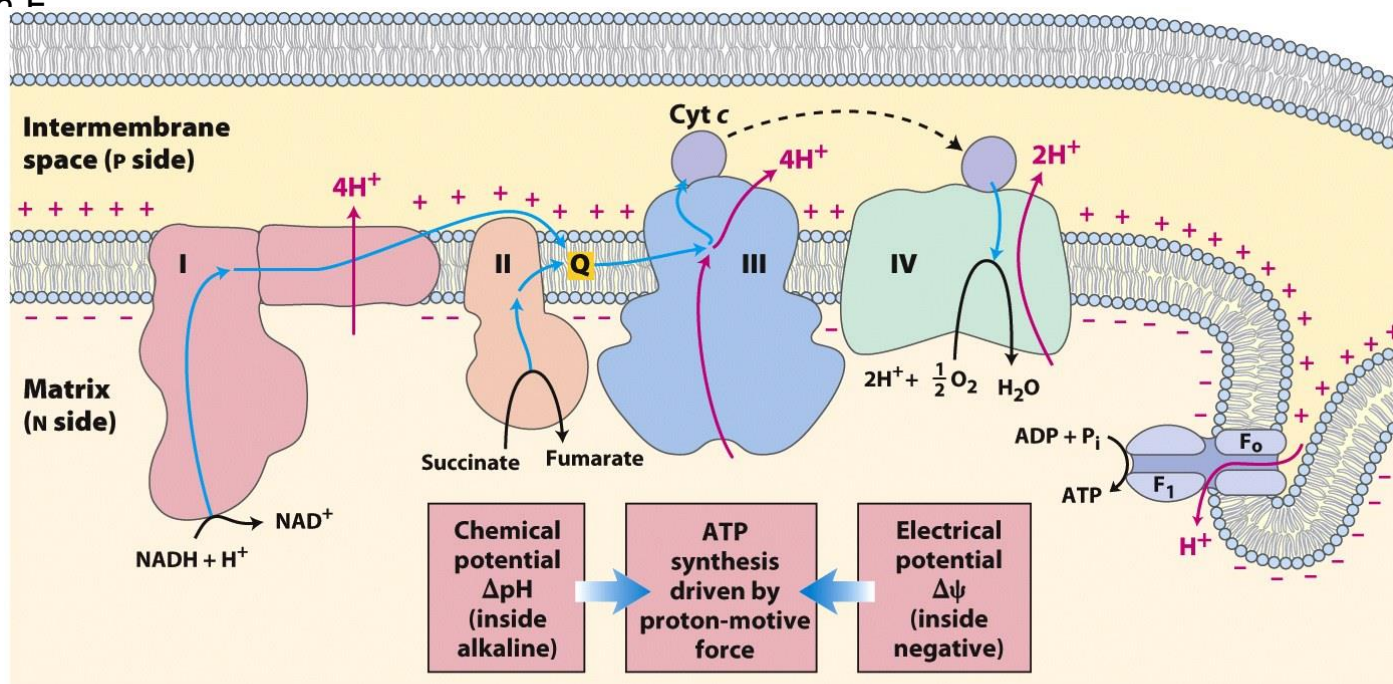


Figure 19-19

Lehninger Principles of Biochemistry, Fifth Edition

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PMF Energized Membrane

Proton-motive force (PMF)

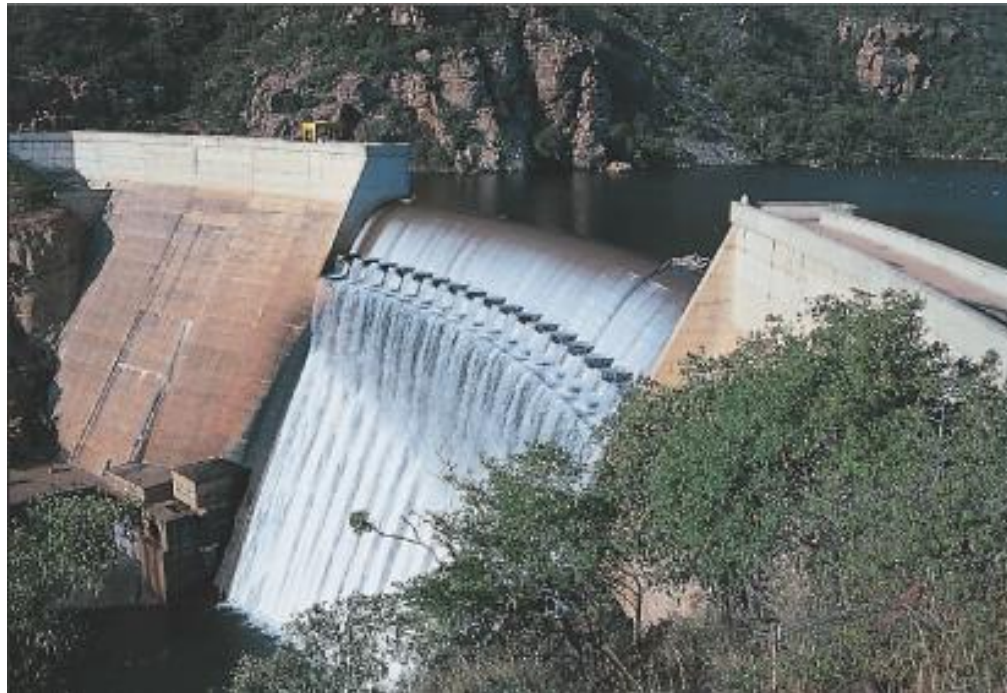


Figure 12-11 Essential Cell Biology, 2/e. (© 2004 Garland Science)

ATP-Synthase/ATPase

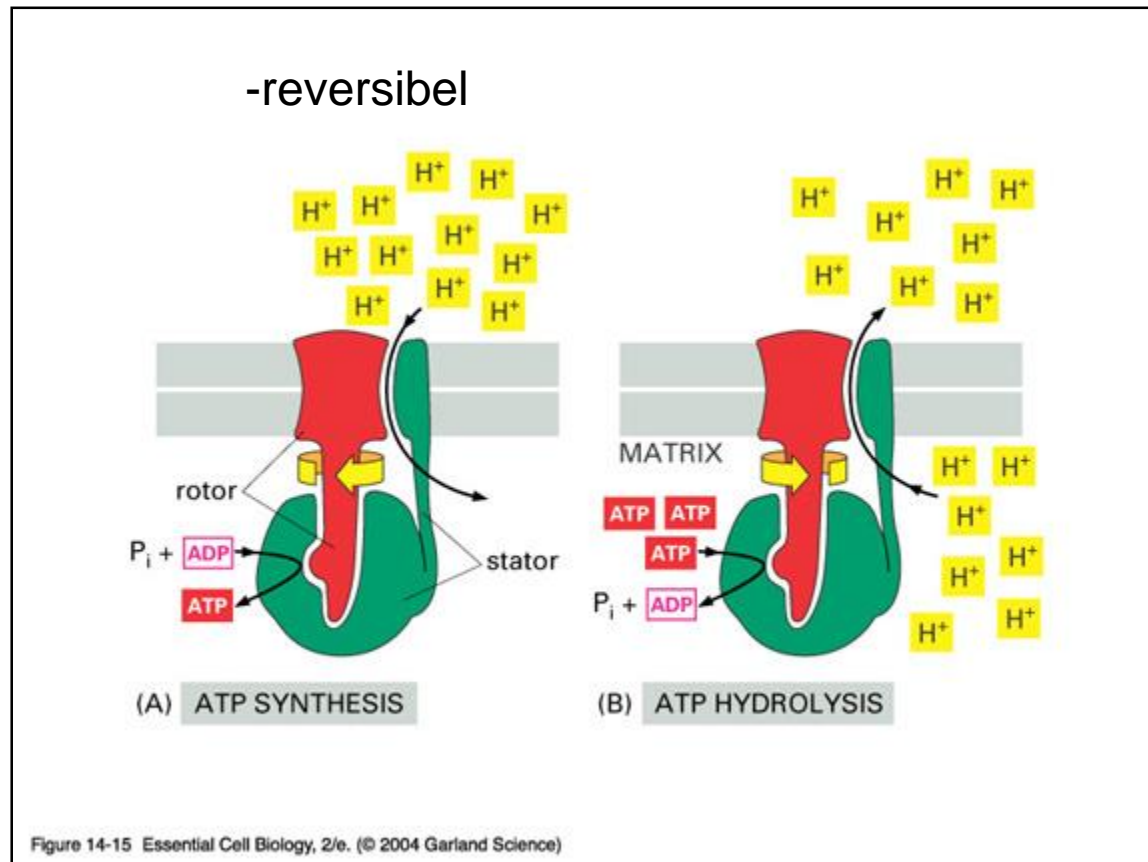
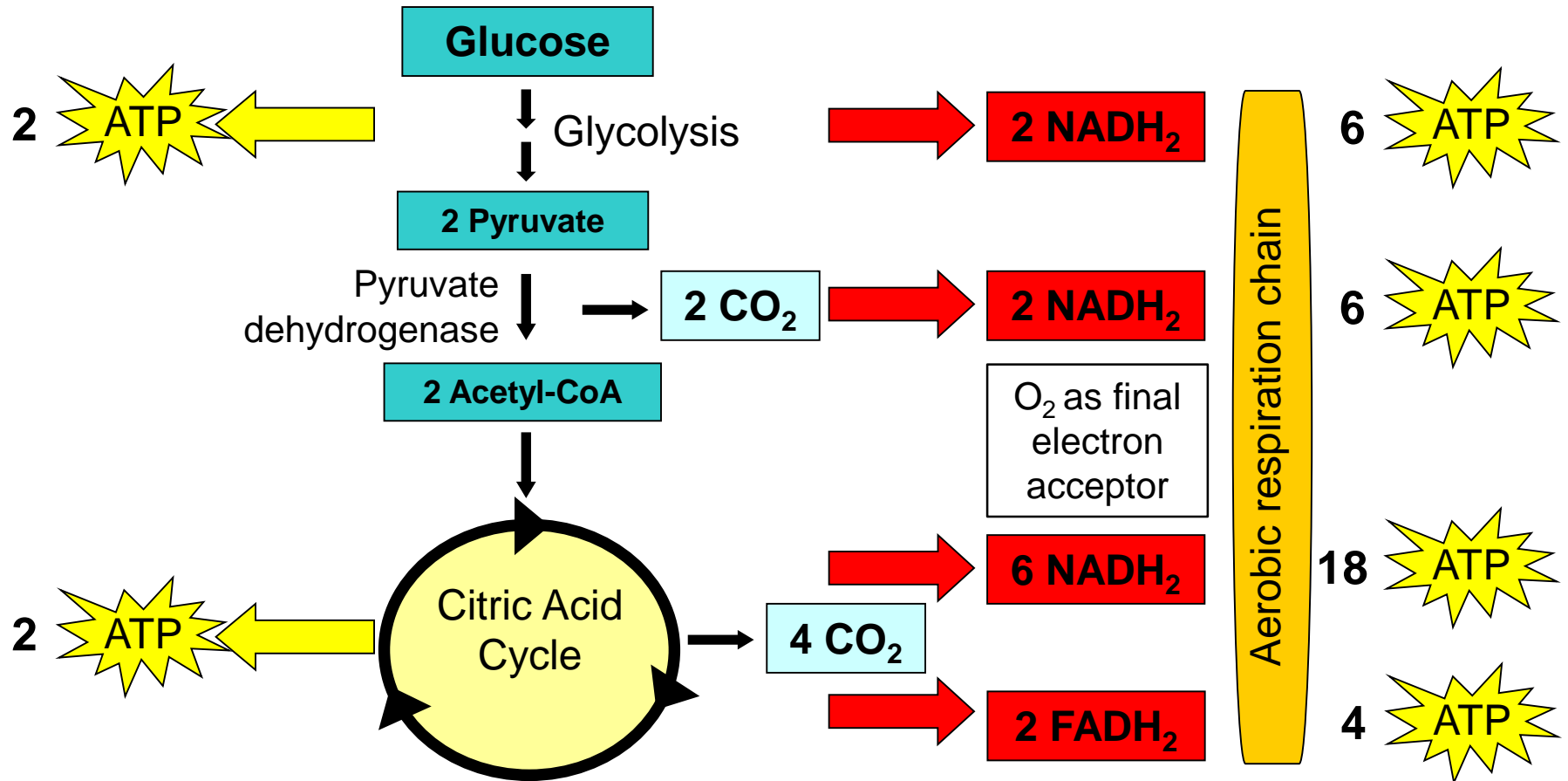


Fig. 14.15 Essential Cell Biology (2nd edition, Alberts, Bray et al.)

Energetics of Carbohydrate Metabolism (Aerobic Respiration)



$$4 \text{ ATP} + 34 \text{ ATP} = 38 \text{ ATP}$$

Aerobic Respiration „Eucaryotes“

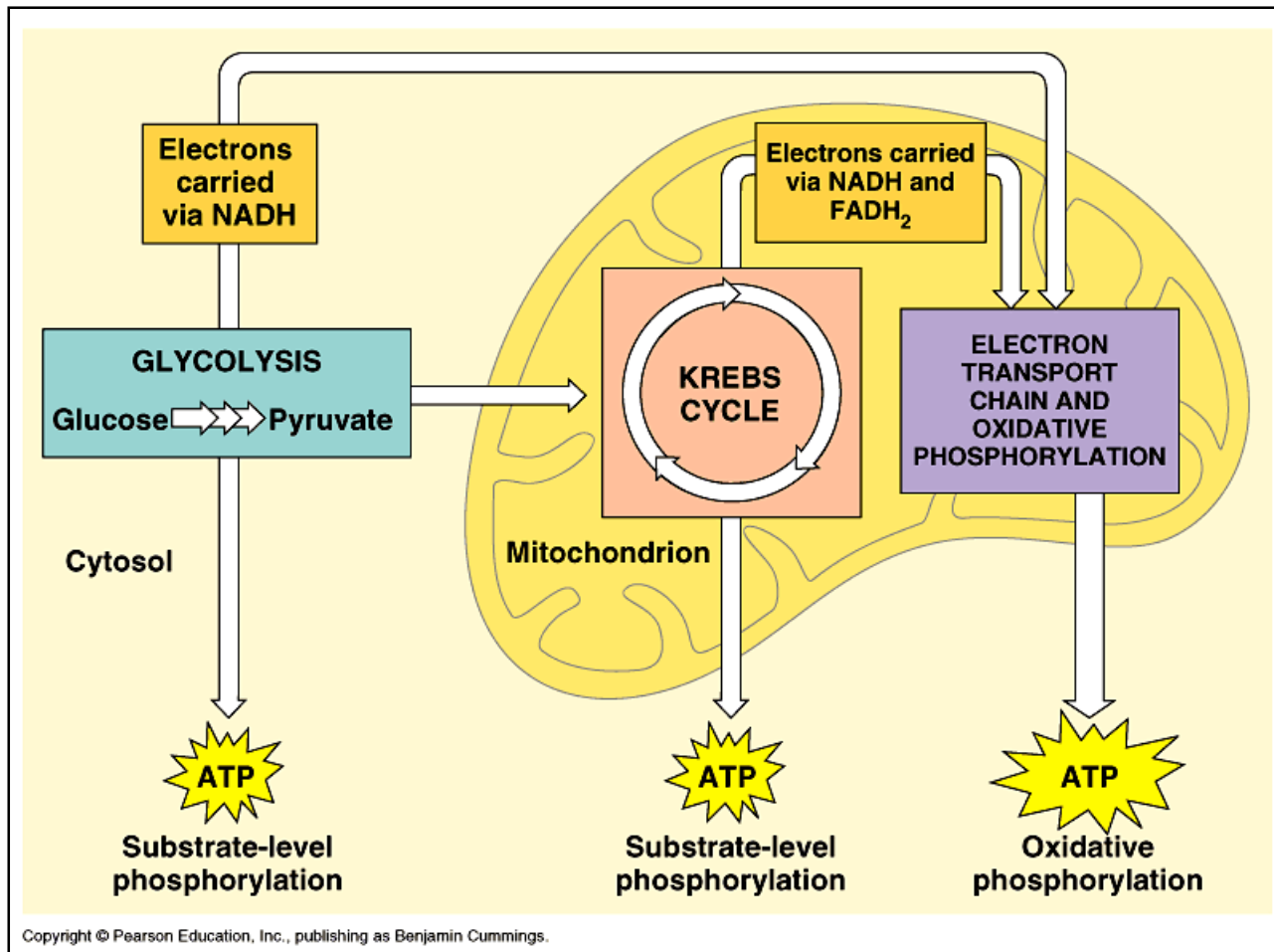


Abb. 9.6 Die Zellatmung im Überblick.
Biologie (Campbell)

Regulation

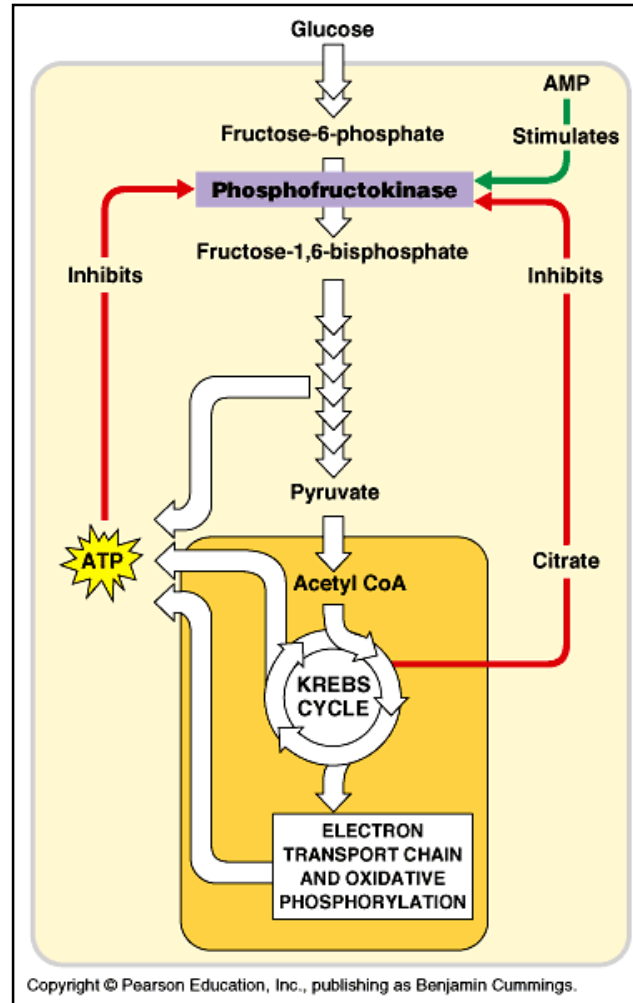


Fig. 9.20 **Die Kontrolle der Zellatmung.**
Biology (6th edition, Campbell & Reece)

Conversion of different Nutrients

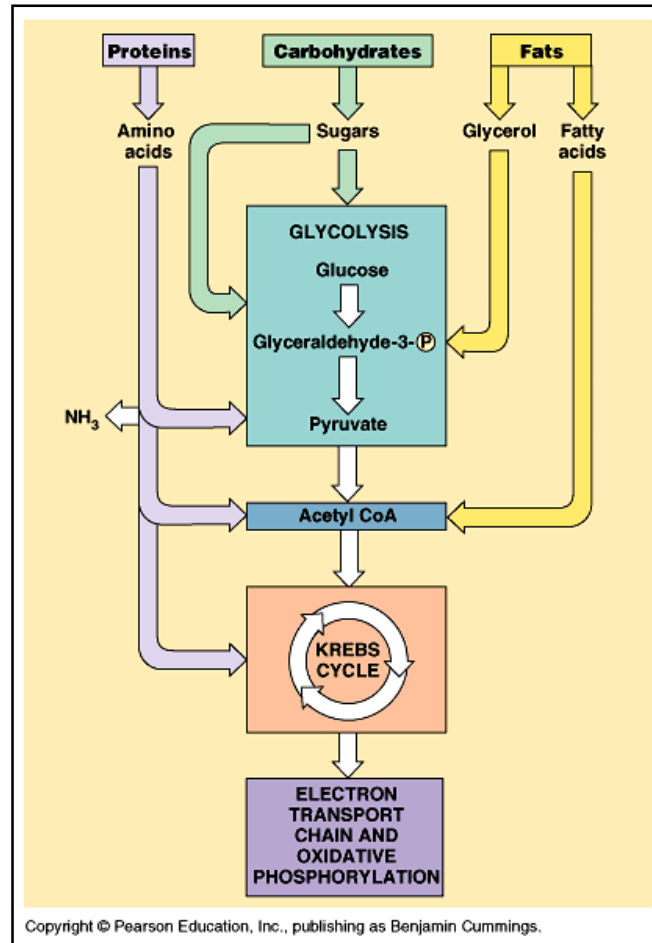


Fig. 9.19 **Catabolism of different nutrients.**
Biology (6th edition, Campbell & Reece)

„Platform Metabolism“ Anabolism

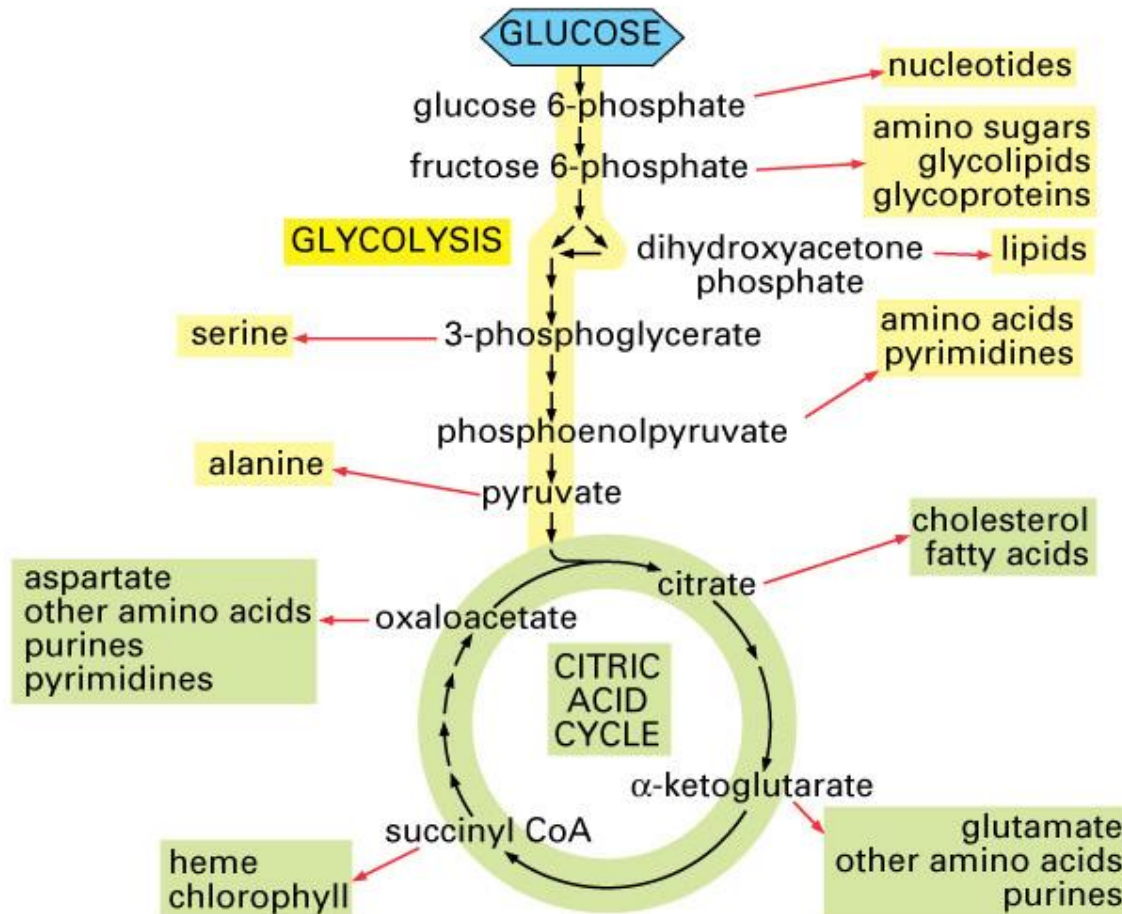


Figure 13-23 Essential Cell Biology, 2/e. (© 2004 Garland Science)

Fermentation



Atmung/Fermentation

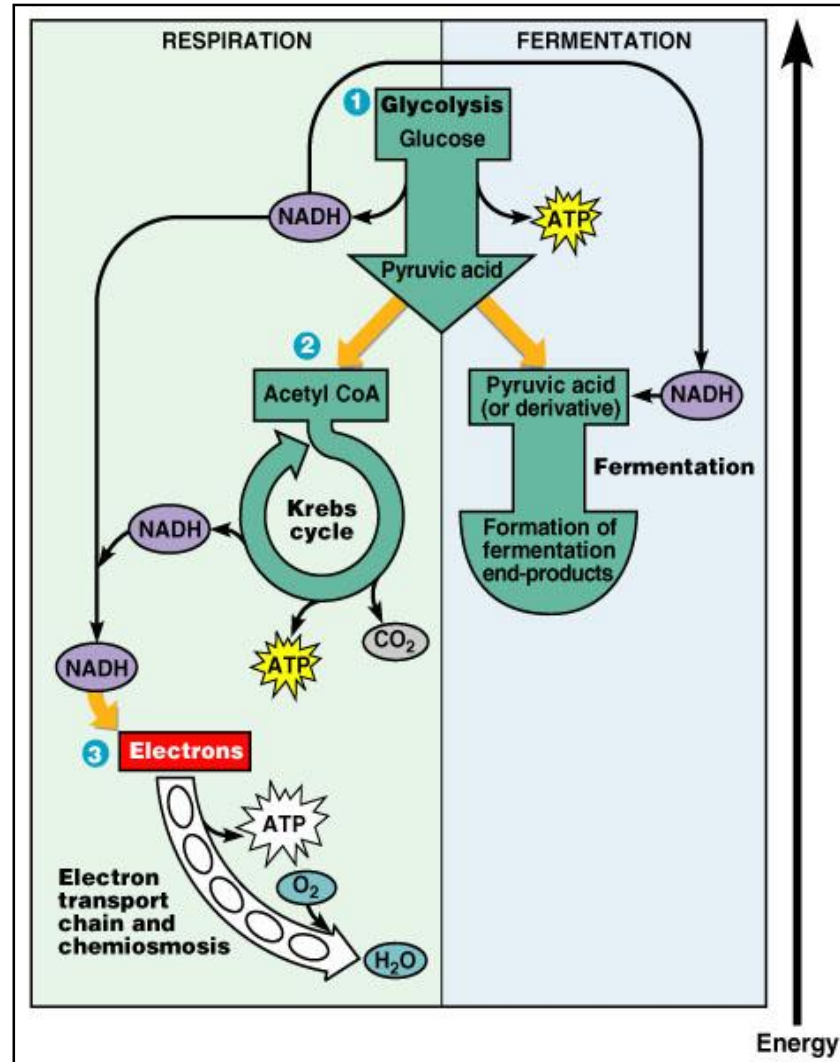
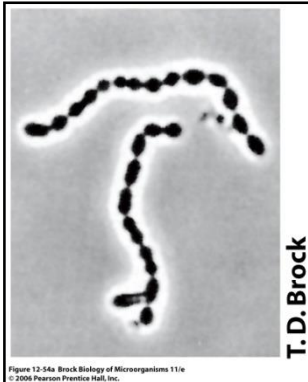


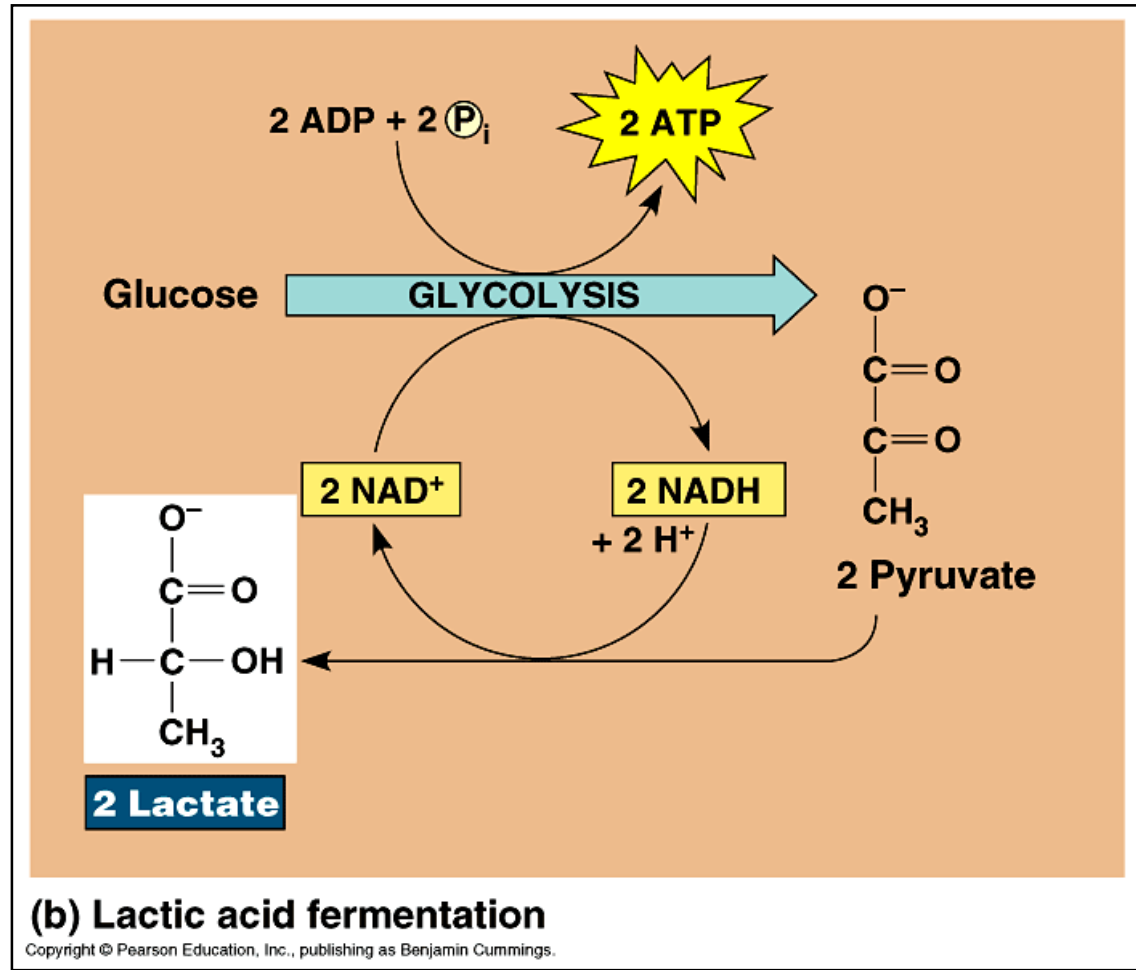
Fig. 5.14 Microbiology: An Introduction (Tortora, Funke, Case)

Lactic Acid Fermentation



T.D. Brock

Lactococcus lactis



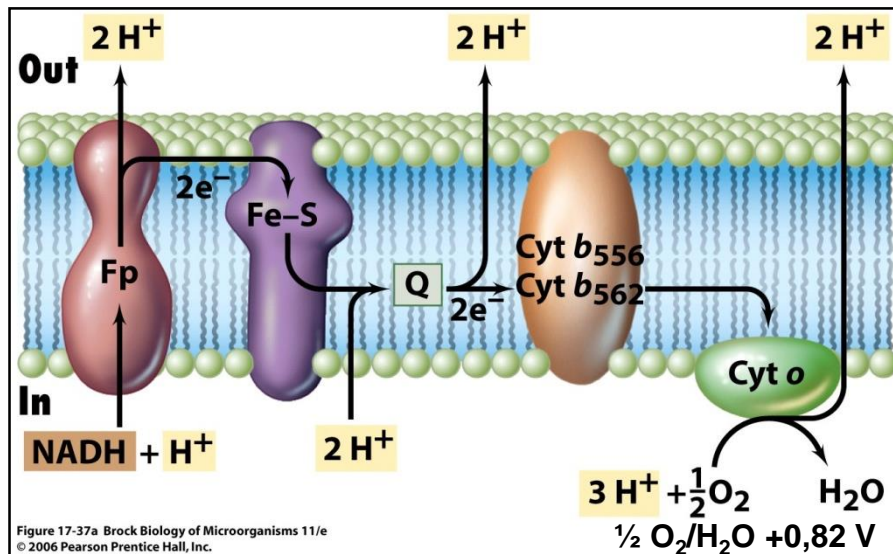
Anaerobic Respiration

- **Alternative electron acceptors in absence of oxygen**
- **Energy source:**
mostly organic compounds (chemoorganotrophic org.);
but also inorganic compounds (chemolithotrophic org.)
- **Electron acceptors:**
Inorganic compounds, NO_3^- , SO_4^{2-} , Fe^{3+} , NO_2^- , S^0 , CO_2
- **Electron transport chain:**
analogue to the aerobic chain (Cytochrome, Quinone, Fe-S Proteine)
- **Facultative aerobes/anaerobes** with aerobic and anaerobic respiration;
- **Obligate anaerobes** only anaerobic respiration

Nitrate Reduction (*E. coli*)

- **Enterobacteriaceae** (e.g. *E. coli*)
- **Facultative anaerobic** Bacteria (anaerobic fermentation)
- Only **reduction of nitrate to nitrite** (nitrate reductase A)

aerob



anaerob (NO_3^-)

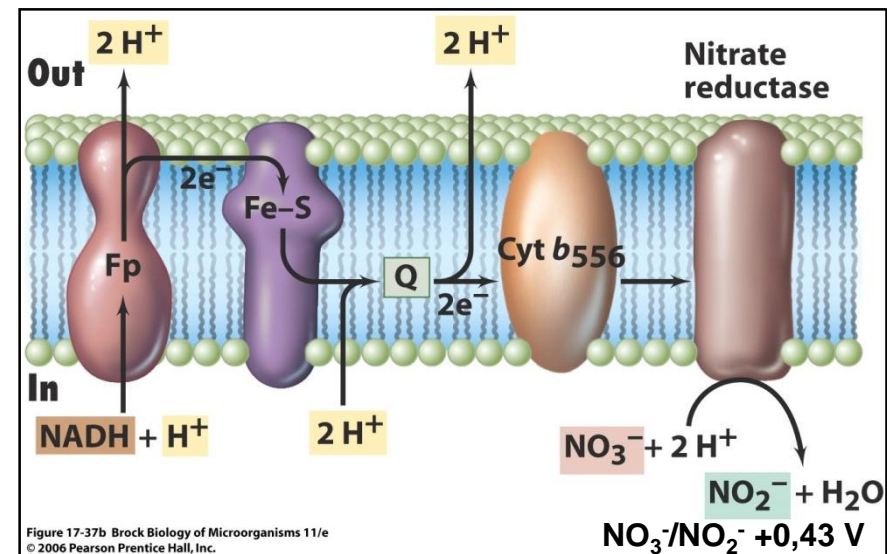
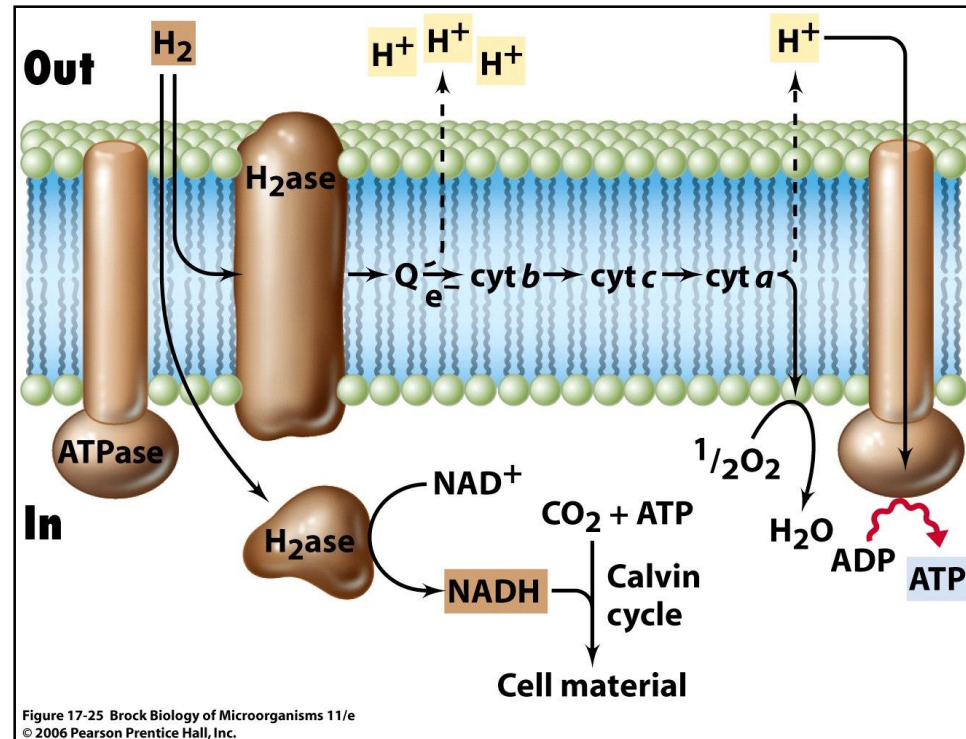


Fig. 17.37 Vergleich aerobe und Nitrat Atmung.
Brock Biology of Microorganisms (10th edition) (Madigan et al.)

Knallgas-Bacteria

- Biologic knallgas-reaktion
„Oxidation of hydrogen“
 $H_2 + \frac{1}{2} O_2 \rightarrow H_2O \quad \Delta G^0' = -237 \text{ kJ}$
„Hydrogenase“
- Different Bacteria:
 - G^- : *Pseudomonas*, *Alcaligenes*, *Paracoccus*,
 - G^+ : *Nocardia*, *Mycobacterium*, *Bacillus*
- Hydrogenase (membrane-bound)
„Electron transport“; some organisms in addition soluble hydrogenase „direct reduction of NAD^+ “
- Chemolithoautotrophe „ CO_2 fixation via calvin cycle“
- Chemoorganotrophic growth (Calvin cycle and hydrogenase repressed)



Photosynthesis

Light- and Dark- Reaction

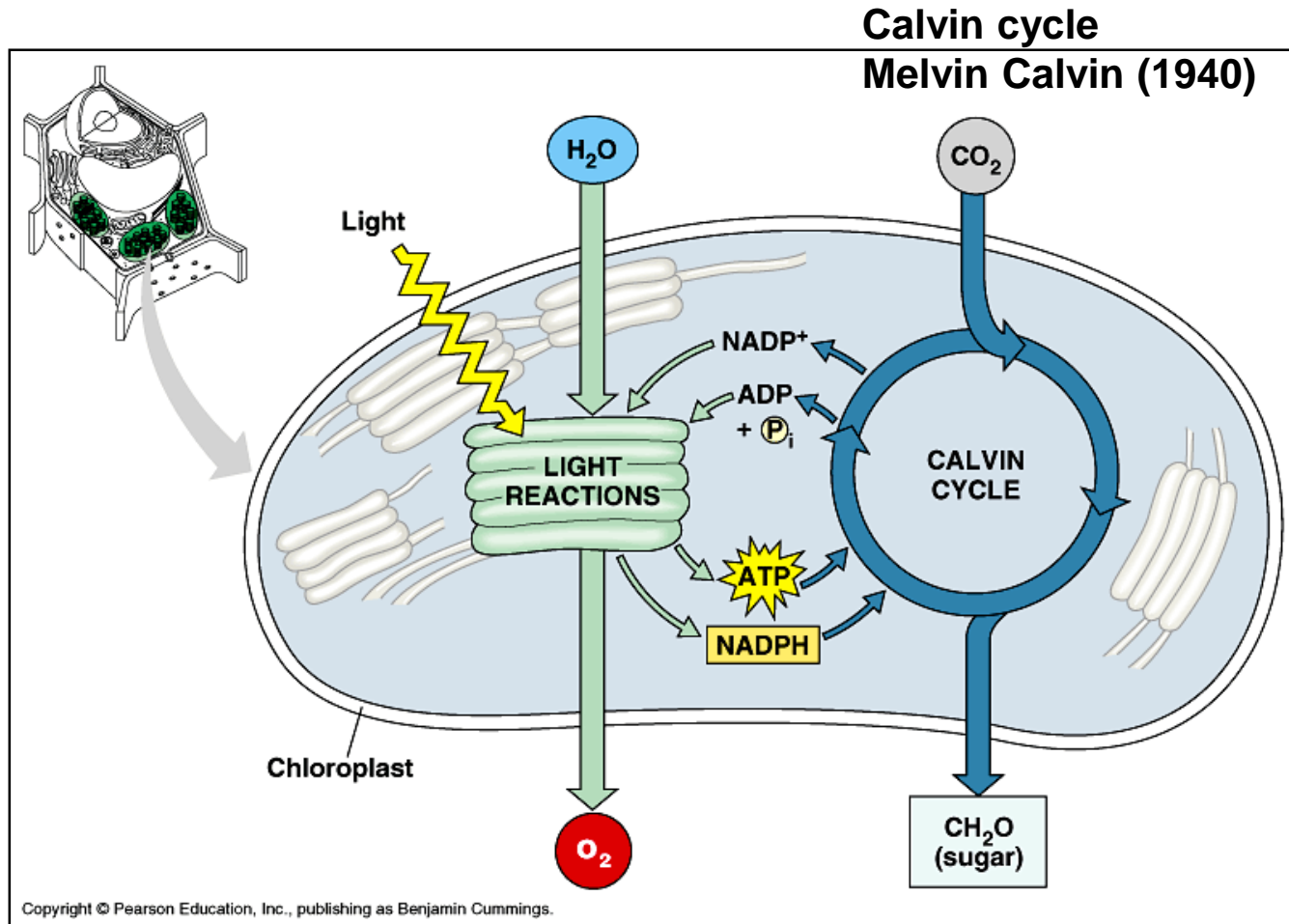


Fig. 10.4 Biology (6th edition, Campbell & Reece)

Non-cyclic Photophosphorylation

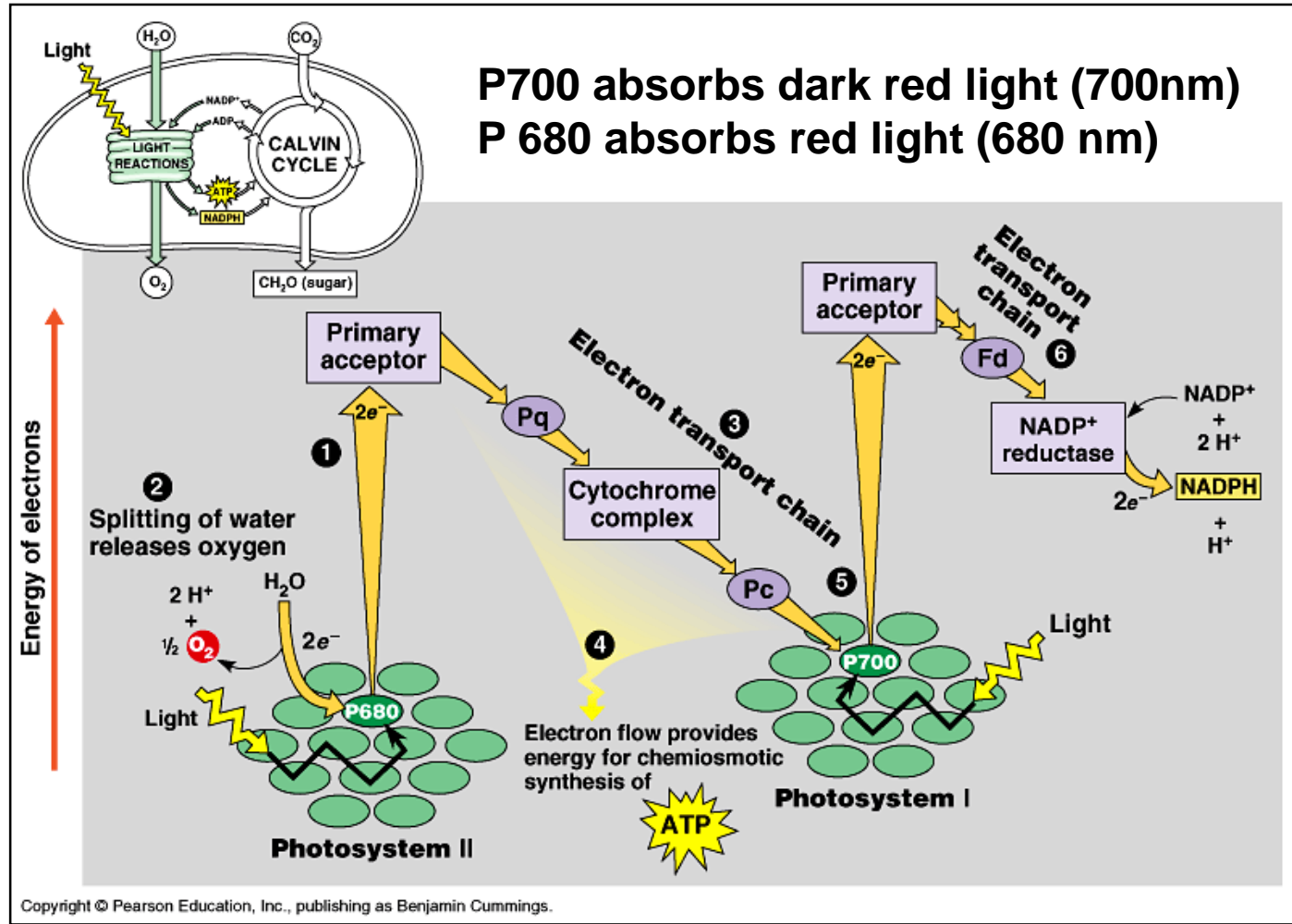


Fig. 10.12 Biology (6th edition, Campbell & Reece)

Plastoquinone (Pq), Cytochrome b₆-f-complex (proton pump), Plastocyanin (Pc, Cu²⁺-Protein)

The Light Reaction and Chemiosmosis

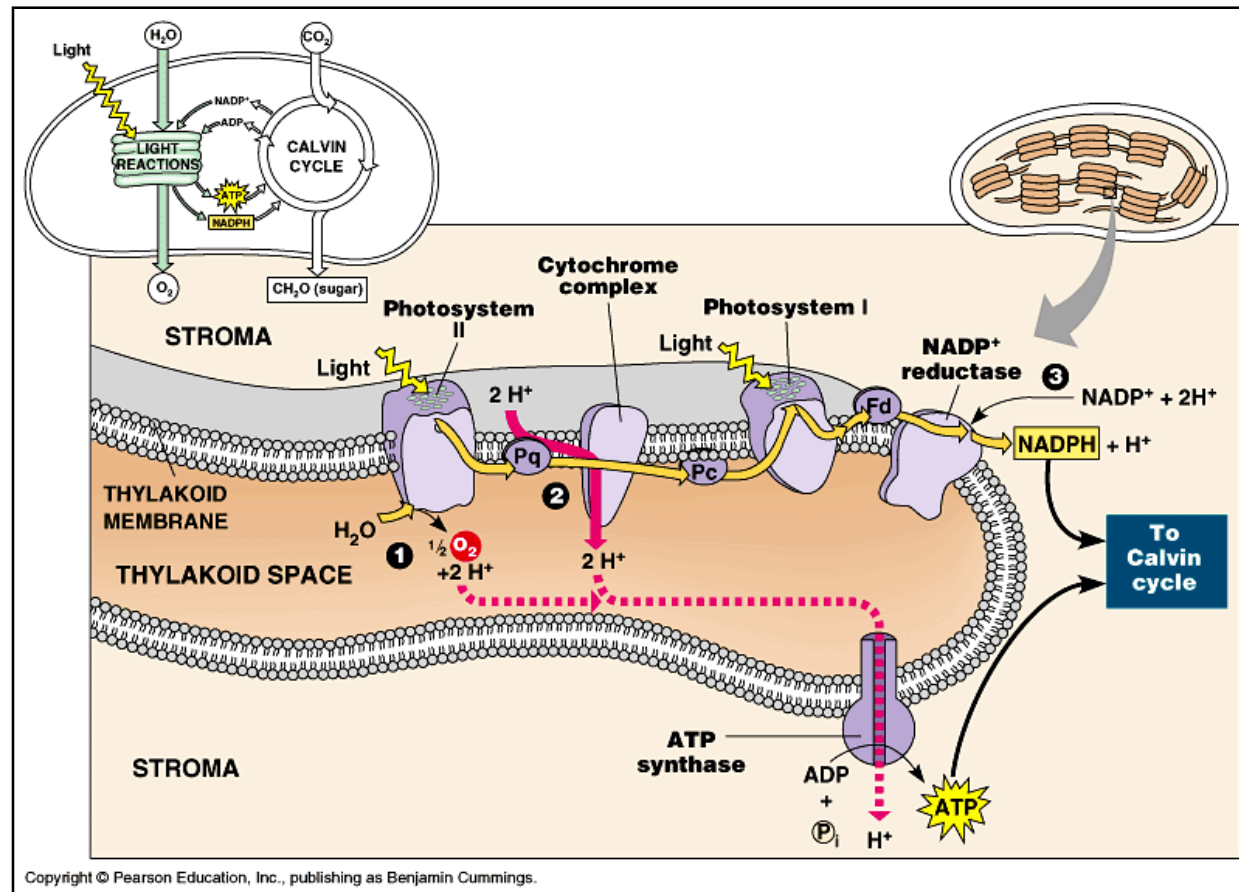


Fig. 10.16 Biology (6th edition, Campbell & Reece)

Comparison of Chemioosmosis in Mitochondrion and Chloroplast

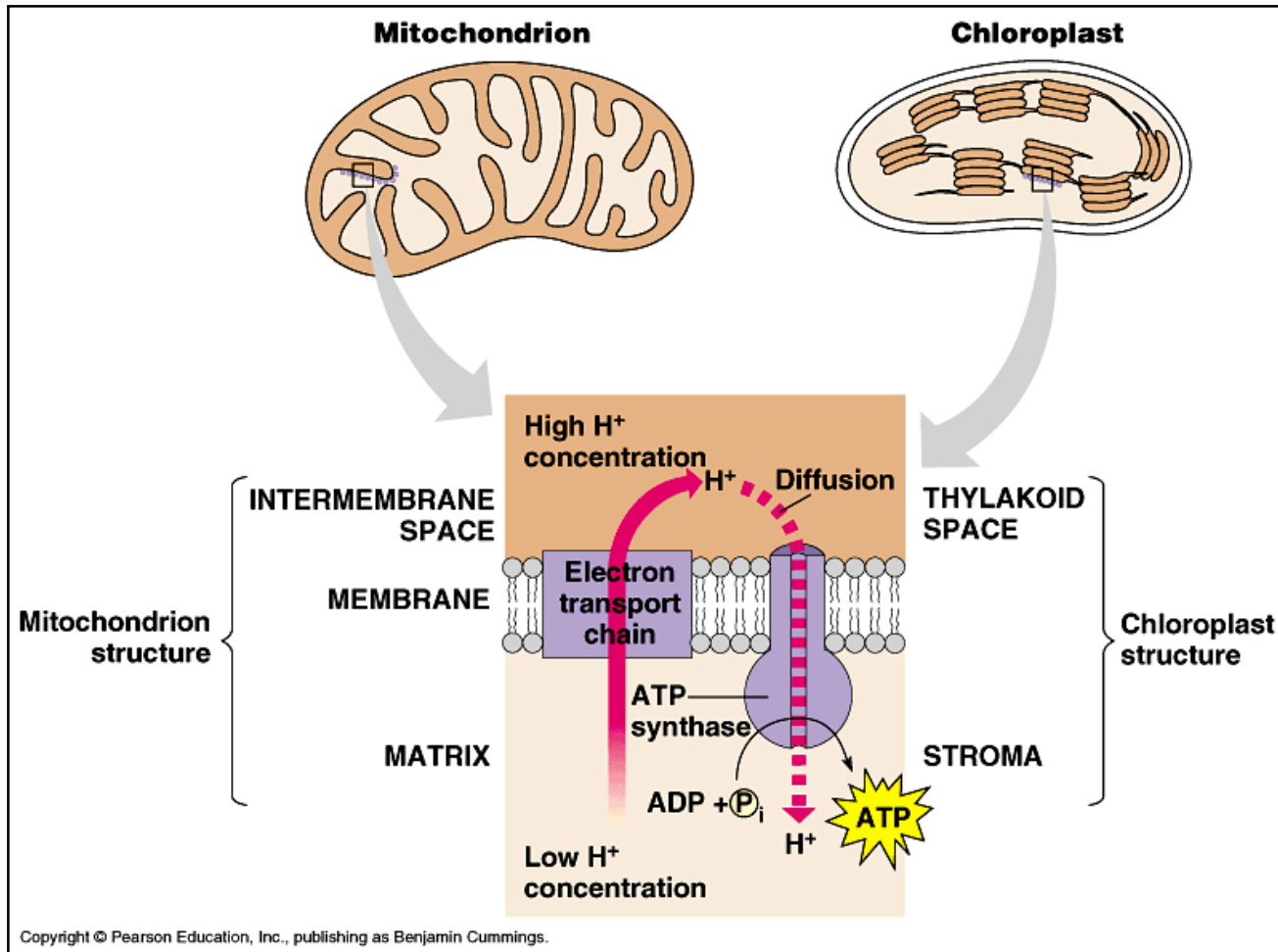


Fig. 10.8 Biology (6th edition, Campbell & Reece)

CO₂ Fixation

CO ₂ fixation pathway	ATP requirement/ pyruvate	Relation to O ₂	Advantages
Calvin cycle	7 ATP	aerobes	Products = sugars, separated from other metabolic pathways
Reductive citric acid cycle	2(-3) ATP	anaerobes, microaerobes	Suited for microaerobic conditions, revers: Oxidation of acetyl-CoA
Reductive acetyl-CoA pathway	1 ATP	strict anaerobes	Suited for the assimilation of C1 units
3-hydroxypropionate pathway	7 ATP	aerobe	Suited for mixotrophic assimilation of fermentation products

The Calvin Cycle „Dark-Reaction“

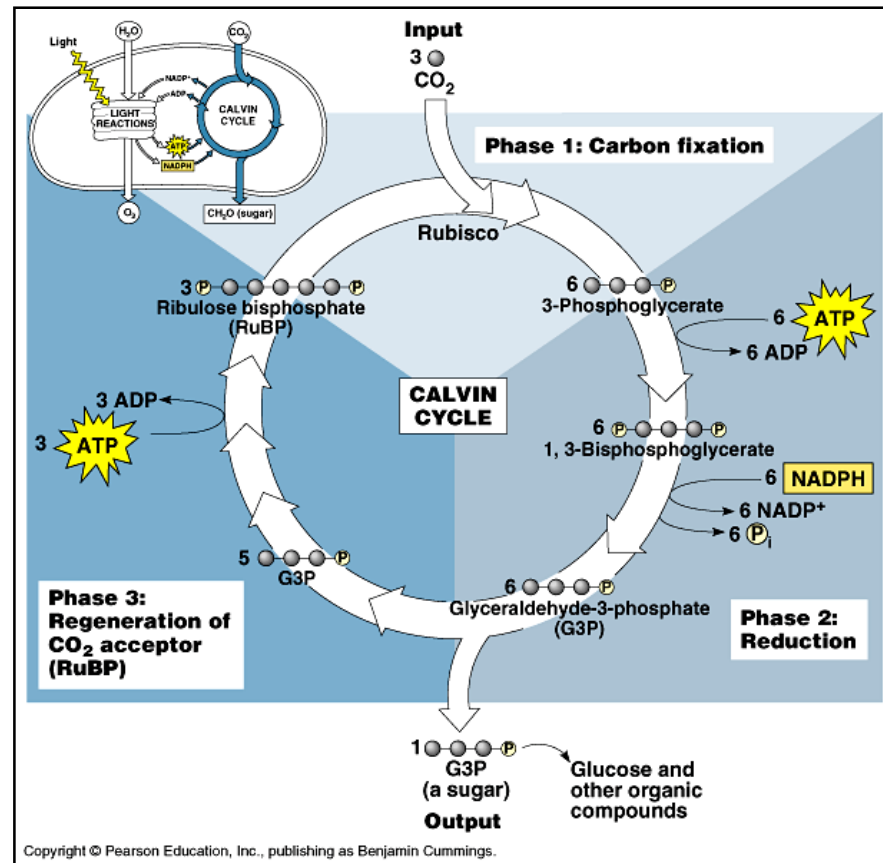


Fig. 10.17 Biology (6th edition, Campbell & Reece)

Key Reactions of the Calvin Cycle (Reductive Pentose Phosphate Cycle)

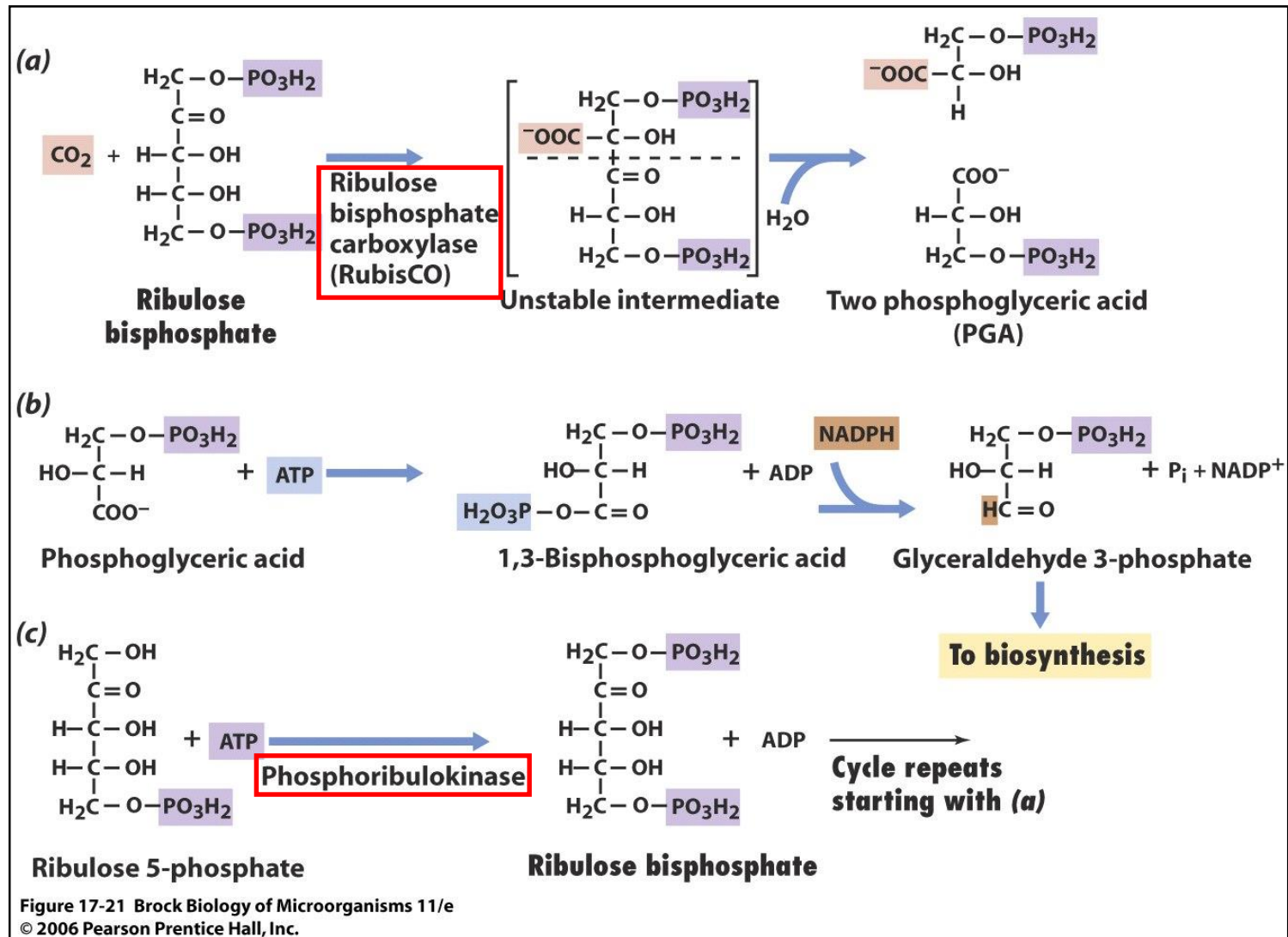
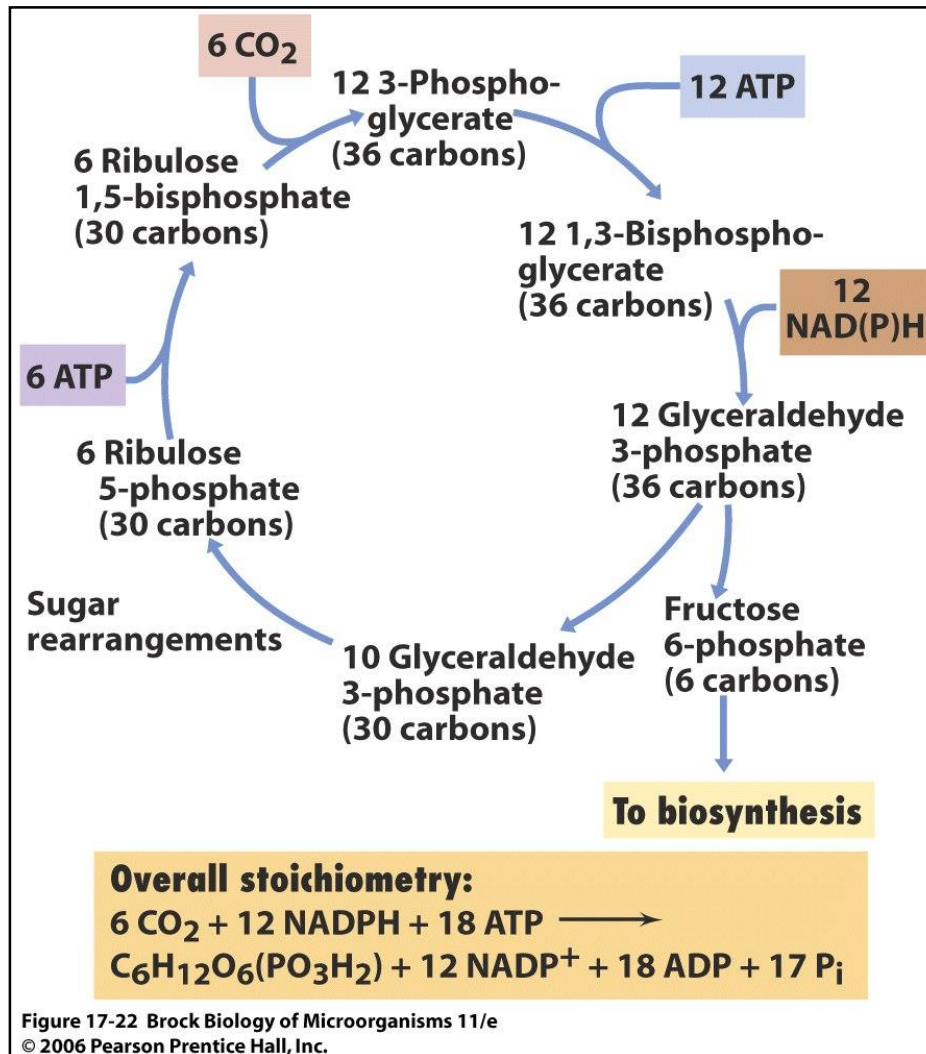


Figure 17-21 Brock Biology of Microorganisms 11/e
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Balance of the Calvin Cycle

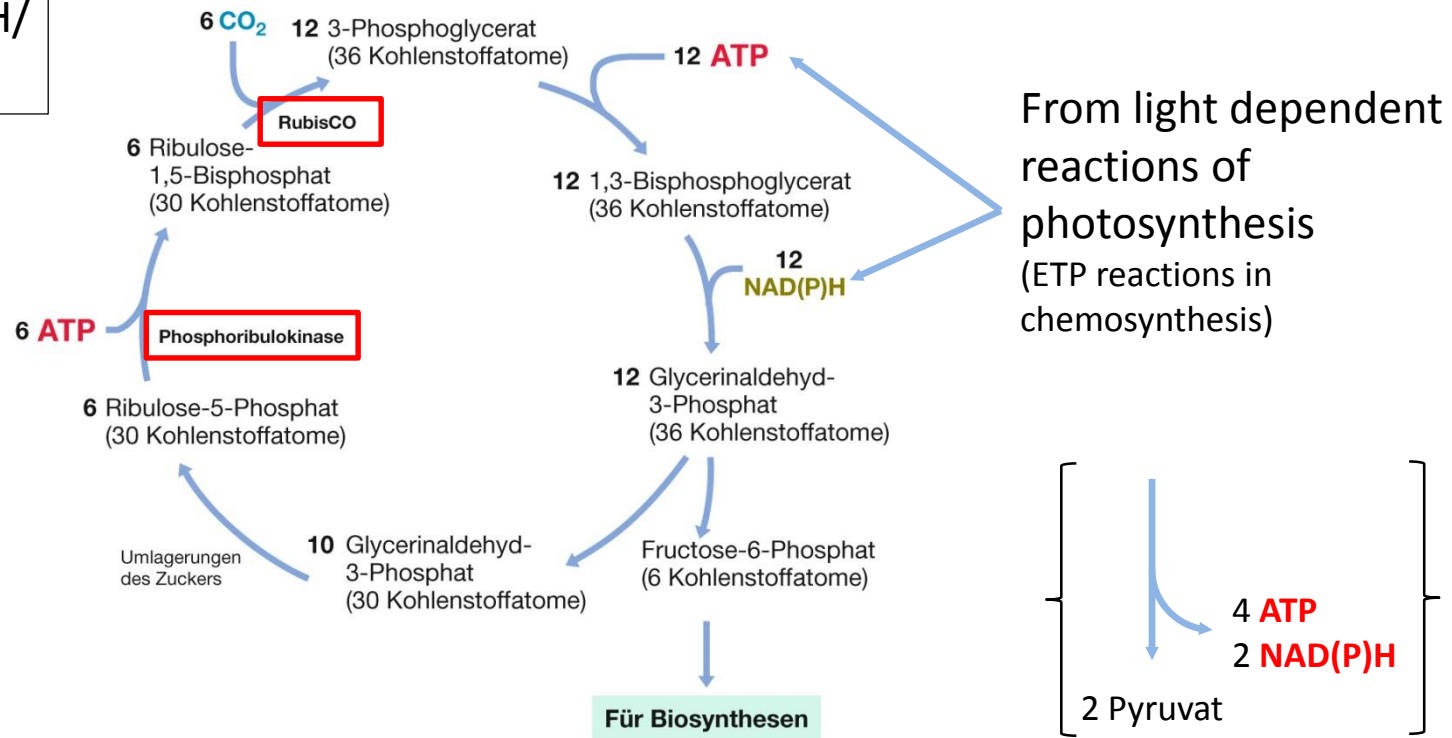


Reductive pentose phosphate cycle

(Calvin-Benson cycle)

RubisCO = Ribulose-bisphosphate carboxylase/oxygenase

Σ 7ATP + 5 NAD(P)H/
pyruvate



- Plants, algae, cyanobacteria, most aerobic and facultativ aerobic Bacteria
- Triosephosphates, 3-phosphoglycerate, sugar phosphates as intermediates

Photosynthesis

Light- and Dark- Reaction

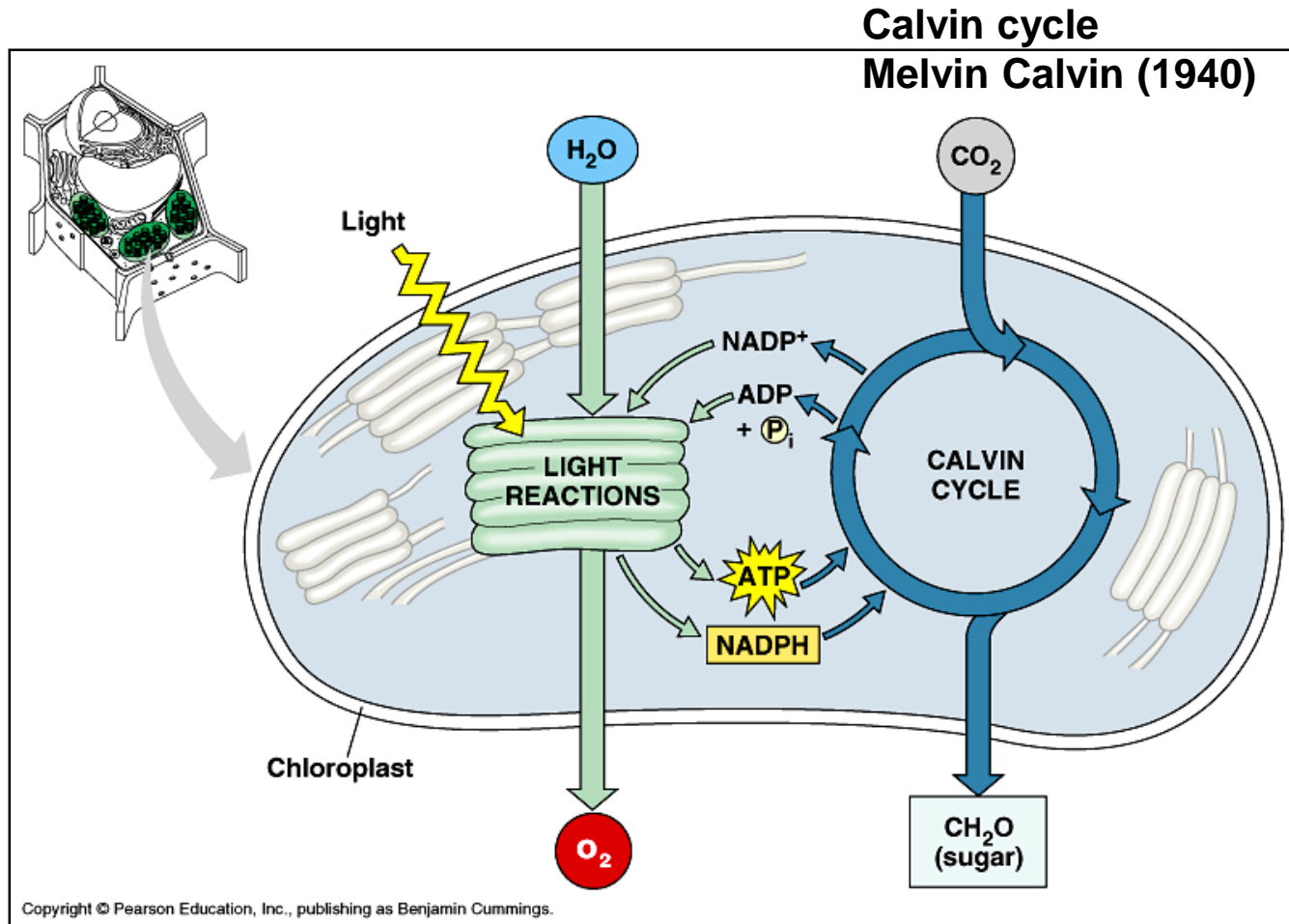
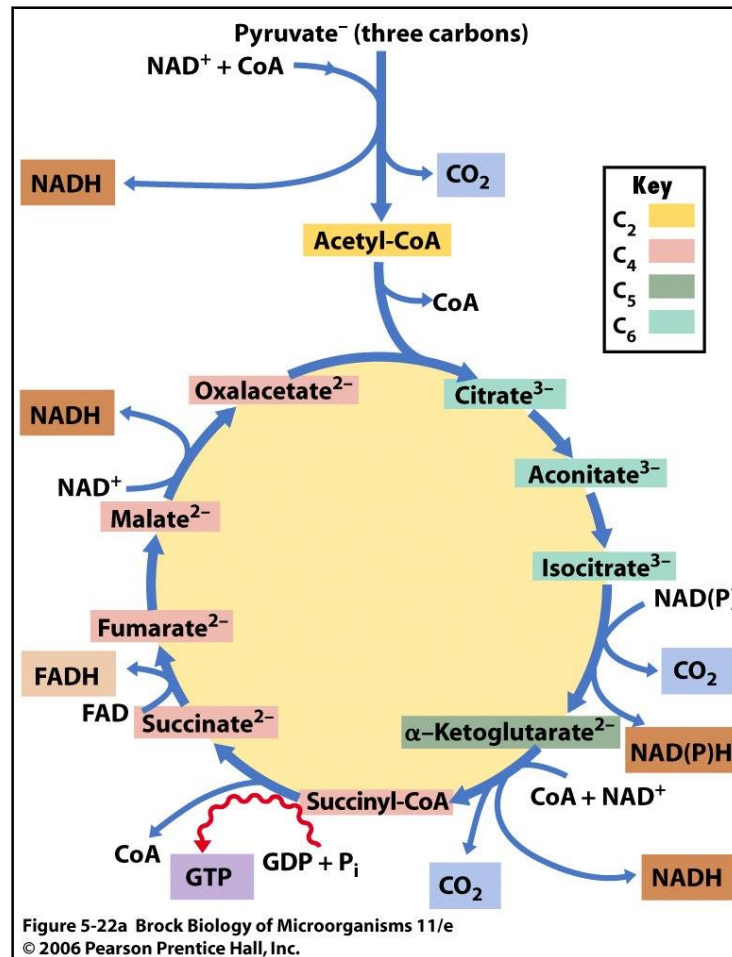


Fig. 10.4 Biology (6th edition, Campbell & Reece)

Pyruvate Dehydrogenase and Oxidative Citric Acid Cycle

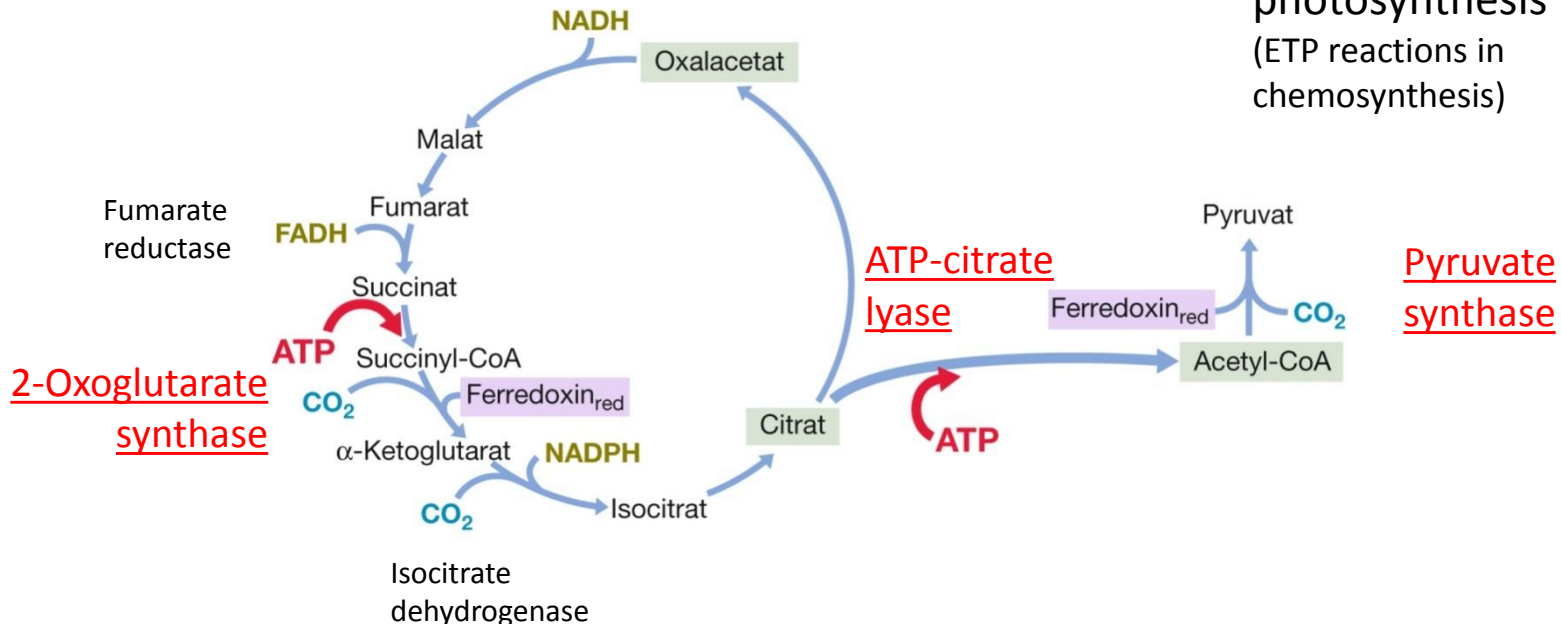


Reductive Citric Acid Cycle

(Arnon-Buchanan cycle)

Σ 2 (-3) ATP + 3 NAD(P)H + 2 ferredoxin/
pyruvate

ATP and ferredoxin from
light dependent
reactions of
photosynthesis
(ETP reactions in
chemosynthesis)



- Anaerobic Green sulfur bacteria (Chlorobiales) and other Proteobacteria, Aquificales (microaerophilic)
- Acetyl-CoA, pyruvate, oxaloacetate, succinyl-CoA, 2-oxoglutarate, (PEP)
- Advantages under anaerobic, microaerophilic conditions

Seminar III

- ATP-Generation via substrate level phosphorylation, electron transport phosphorylation
- (Eukaryotic/prokaryotic cell)
- Bacterial cell wall
- Antibiotic resistance

Prokaryotes & Eukaryotes

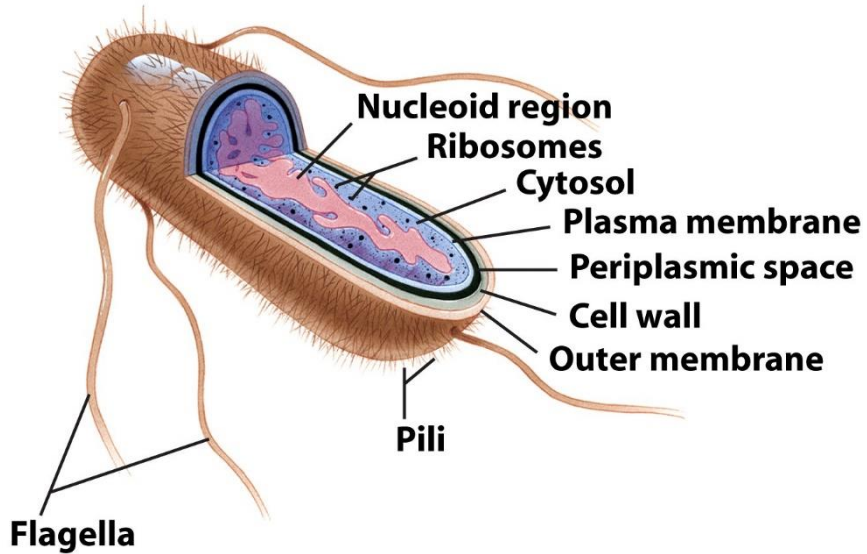


Figure 1-14 Principles of Biochemistry, 4/e
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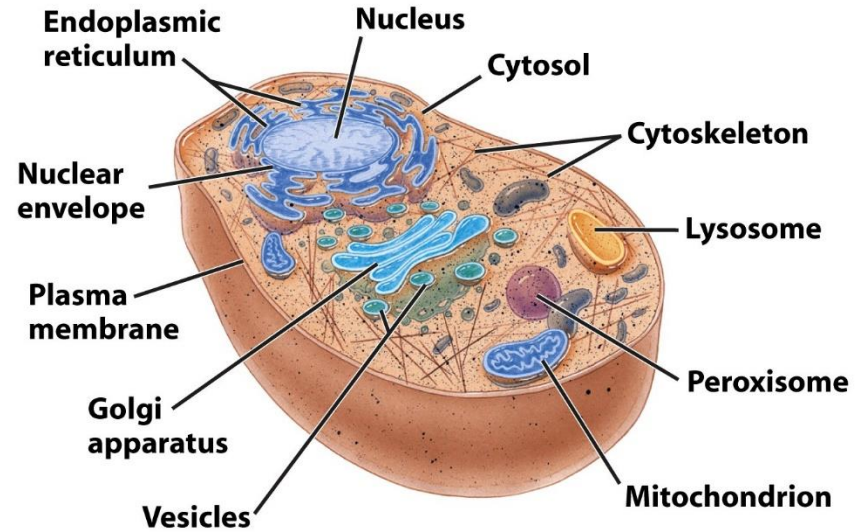
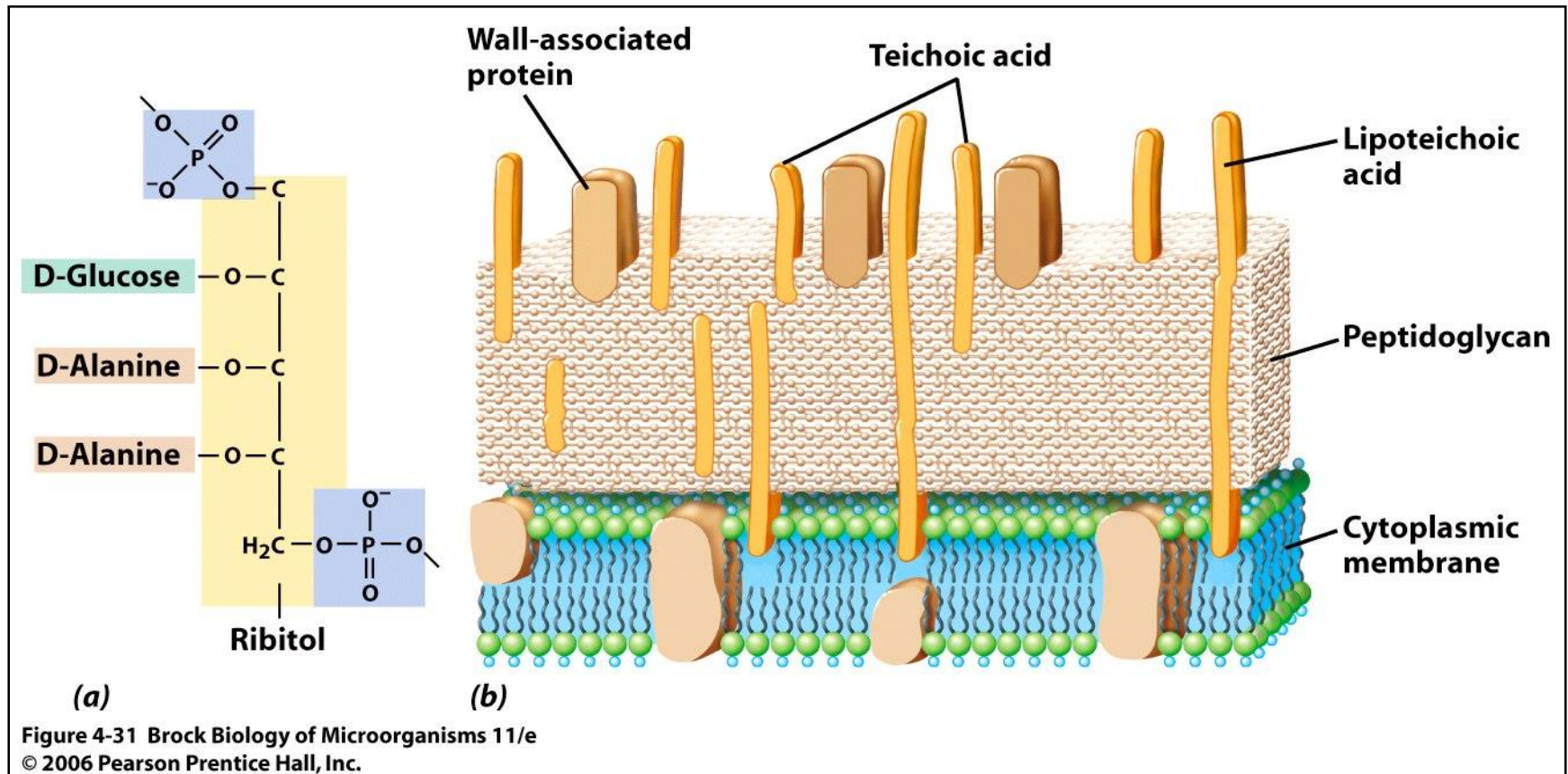


Figure 1-15a Principles of Biochemistry, 4/e
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Murein: Cell wall (Bacteria)

- Gram positive cell wall



Murein: Cell wall (Bacteria)

- Gram negative cell wall

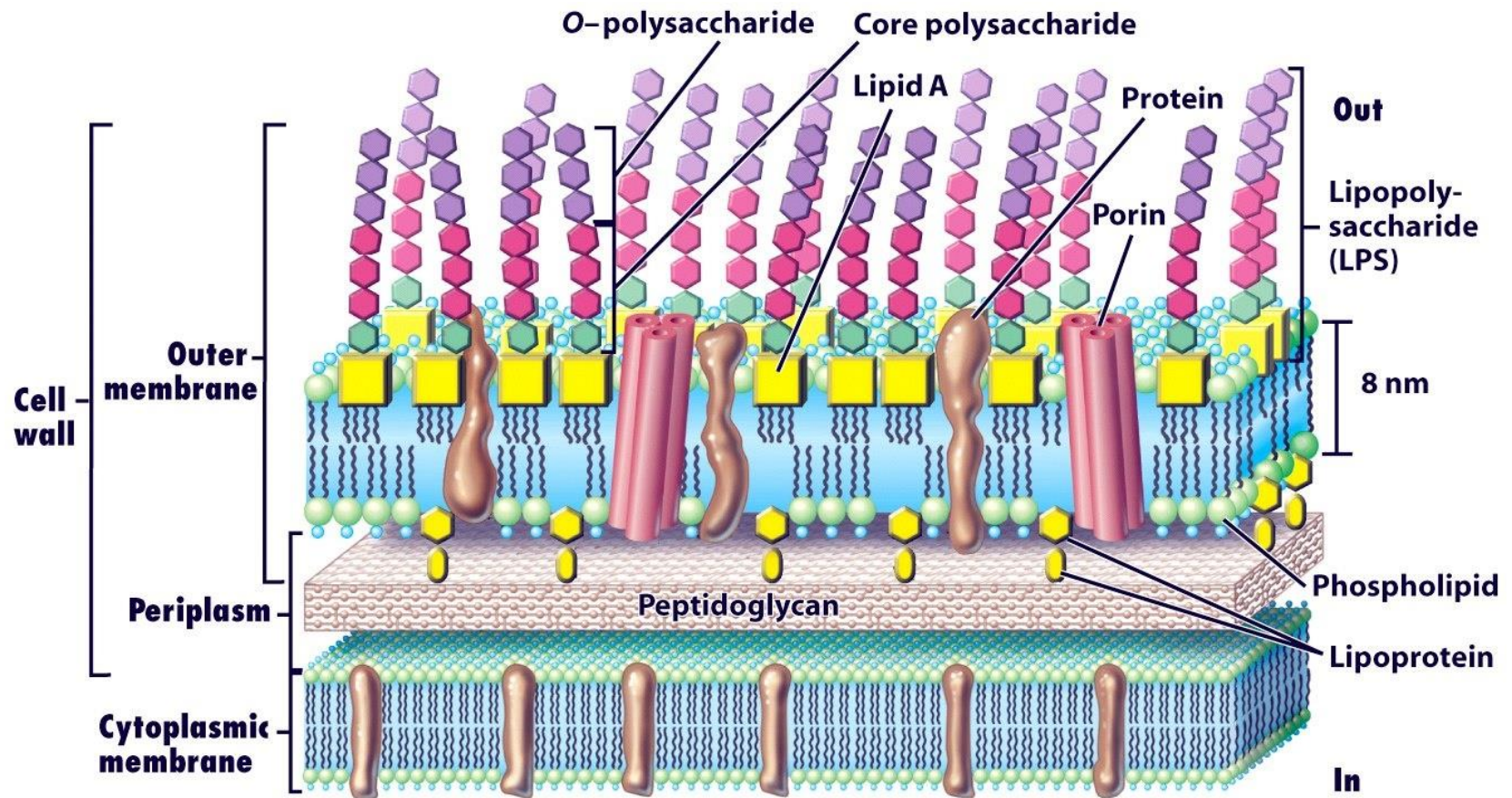


Figure 4-35a Brock Biology of Microorganisms 11/e
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N-acetylglucosamine

- N-acetylglucosamine, a sugar derivative, basic building block for **chitin** and **murein**.



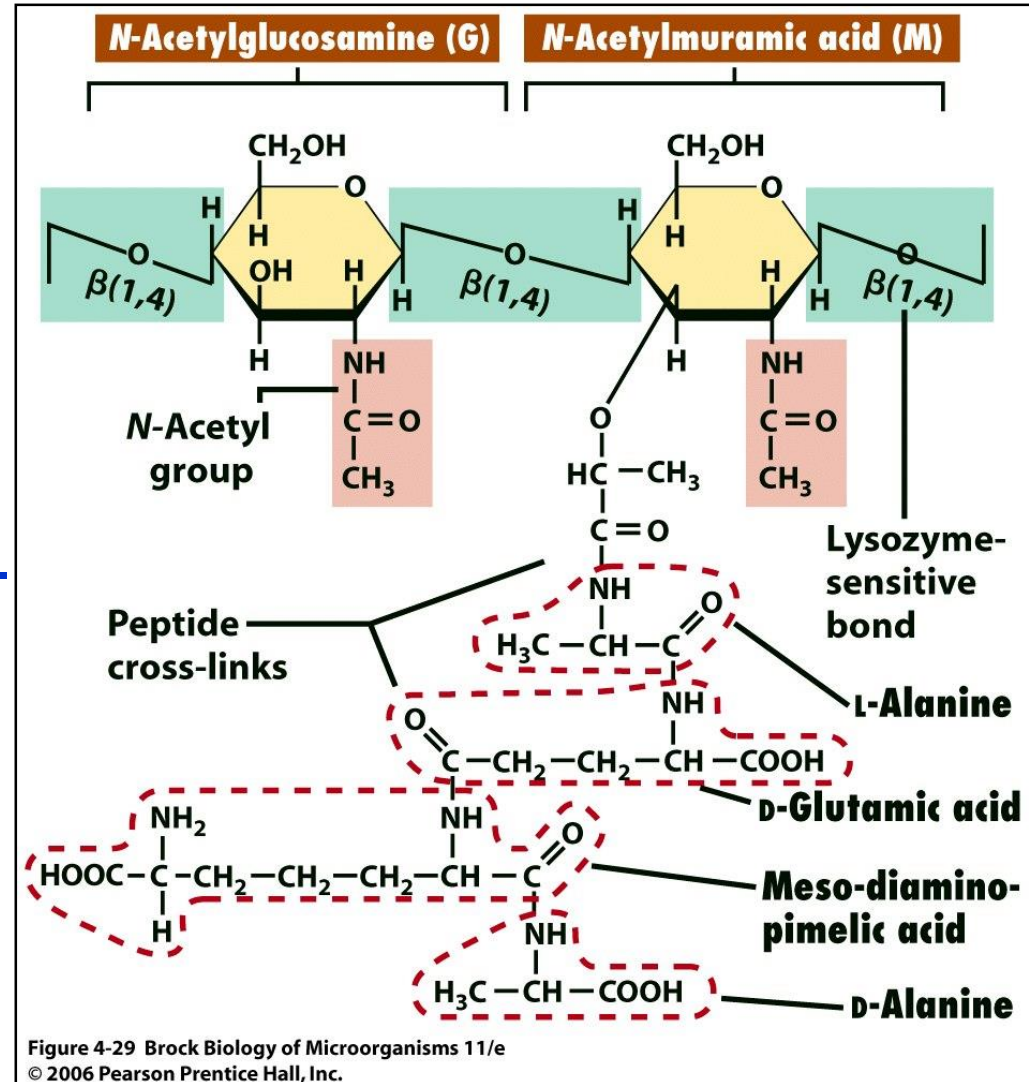
Murein: Cell wall (Bacteria)

Structure of the polysaccharide in bacterial cell wall **peptidoglycan**.

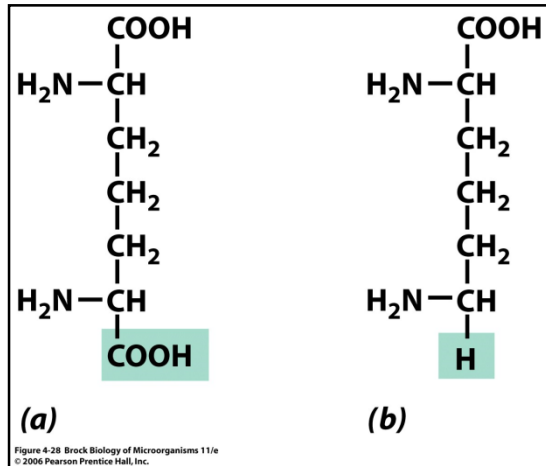
The glycan is a polymer of alternating **GlcNAc** and **N-acetylmuramic acid** (MurNAc, Lactic acid linked to C-4 atom) residues.

Alternating **peptide chains of D- and L-amino acids**

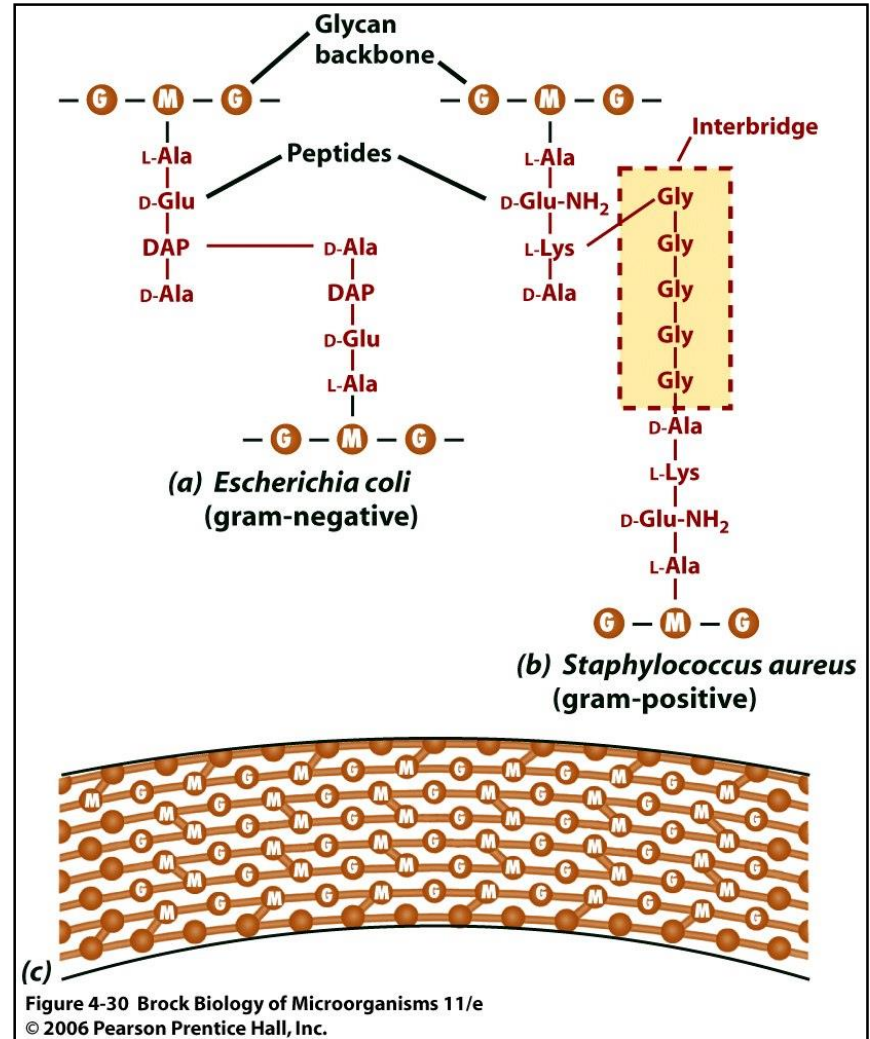
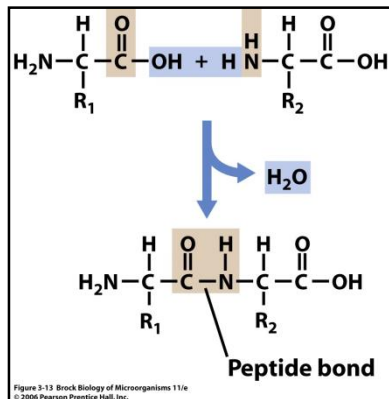
Linkage of the L-alanine amino group (amide linkage) with the lactylcarboxylgroup of a MurNac residue



Murein: Cell wall (Bacteria)



(a) Diaminopimelinsäure
(b) Lysin



Murein: Cell wall (Bacteria)

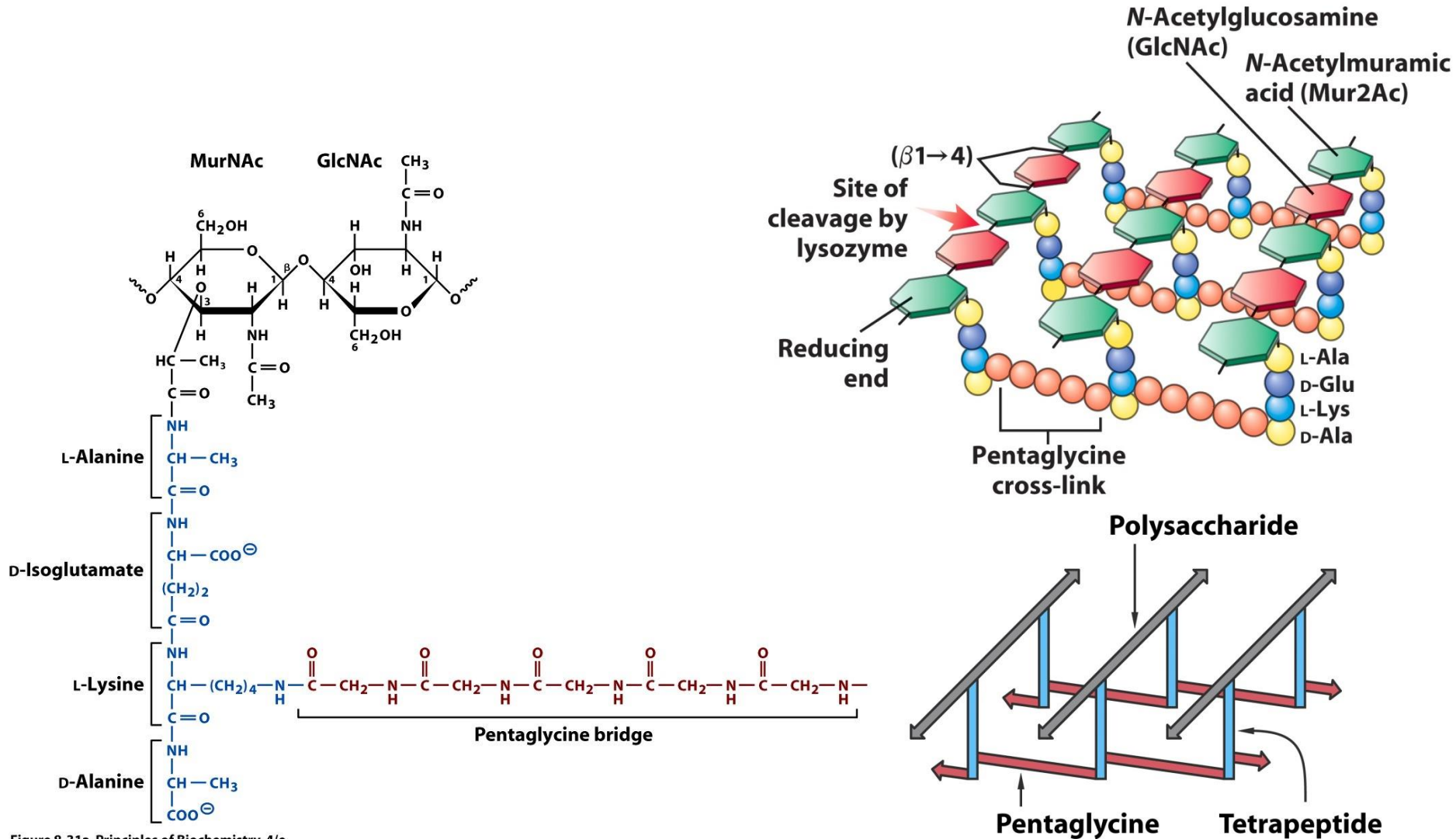


Figure 8-31a Principles of Biochemistry, 4/e
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Figure 8-31b Principles of Biochemistry, 4/e
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Glycoconjugate

- Glycolipids
 - Lipopolysaccharide (Gram negative Bacteria, outer membrane)

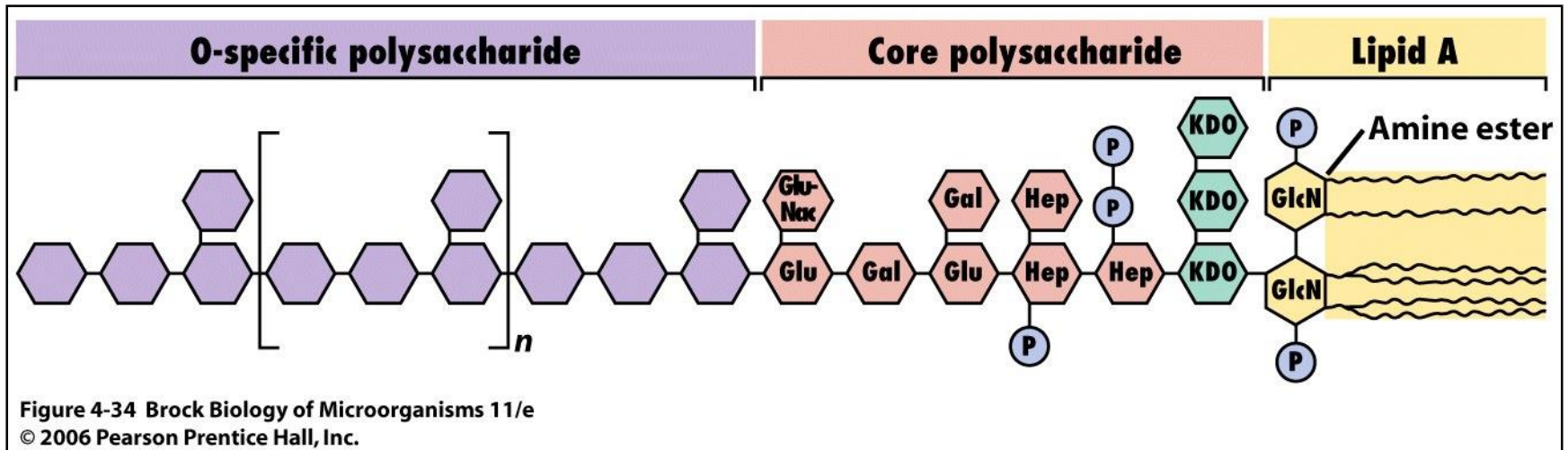


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Mode of Action of Some Major Antimicrobial Agents

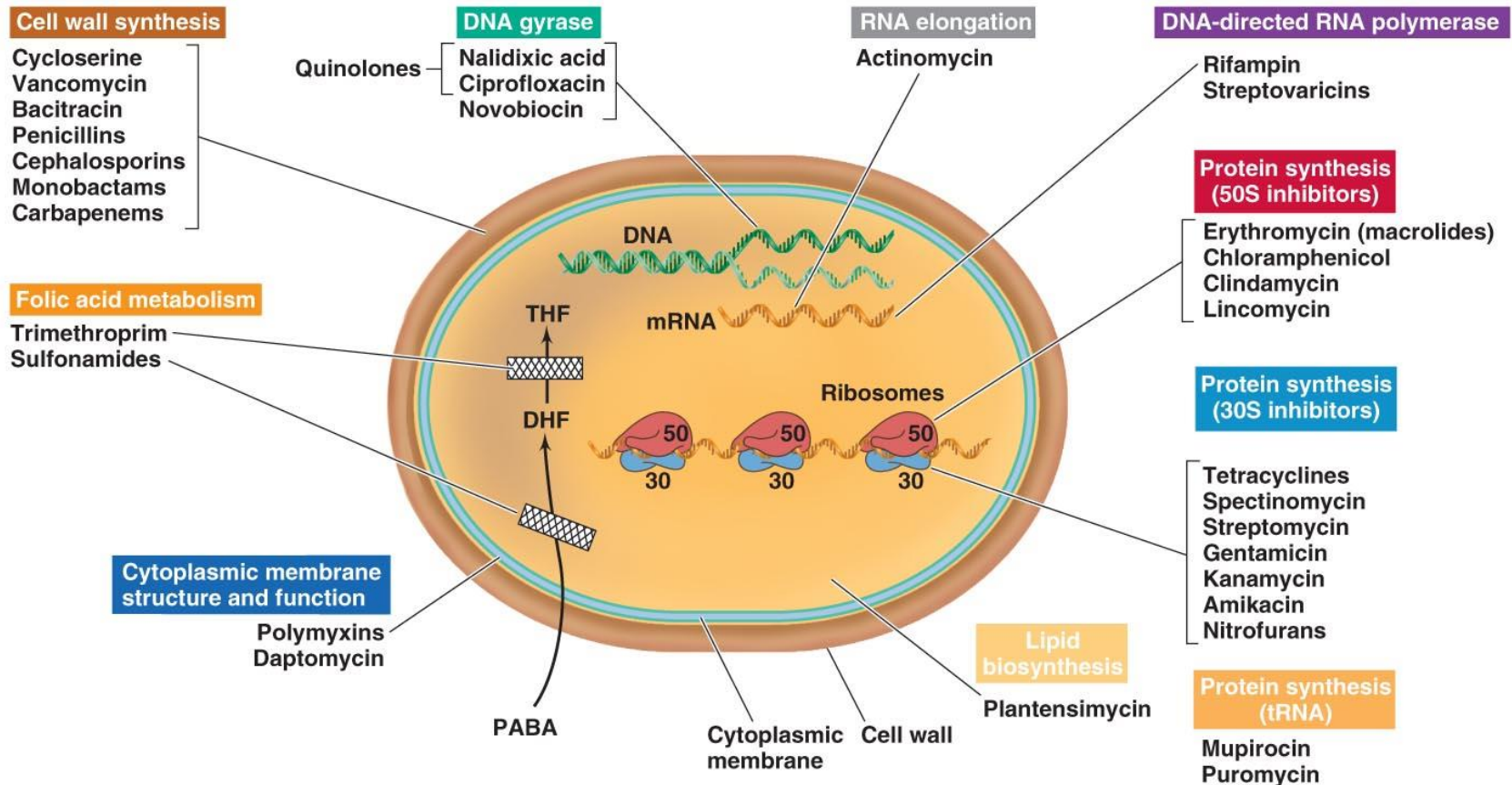


Figure 27.12

Antimicrobial Spectrum of Activity

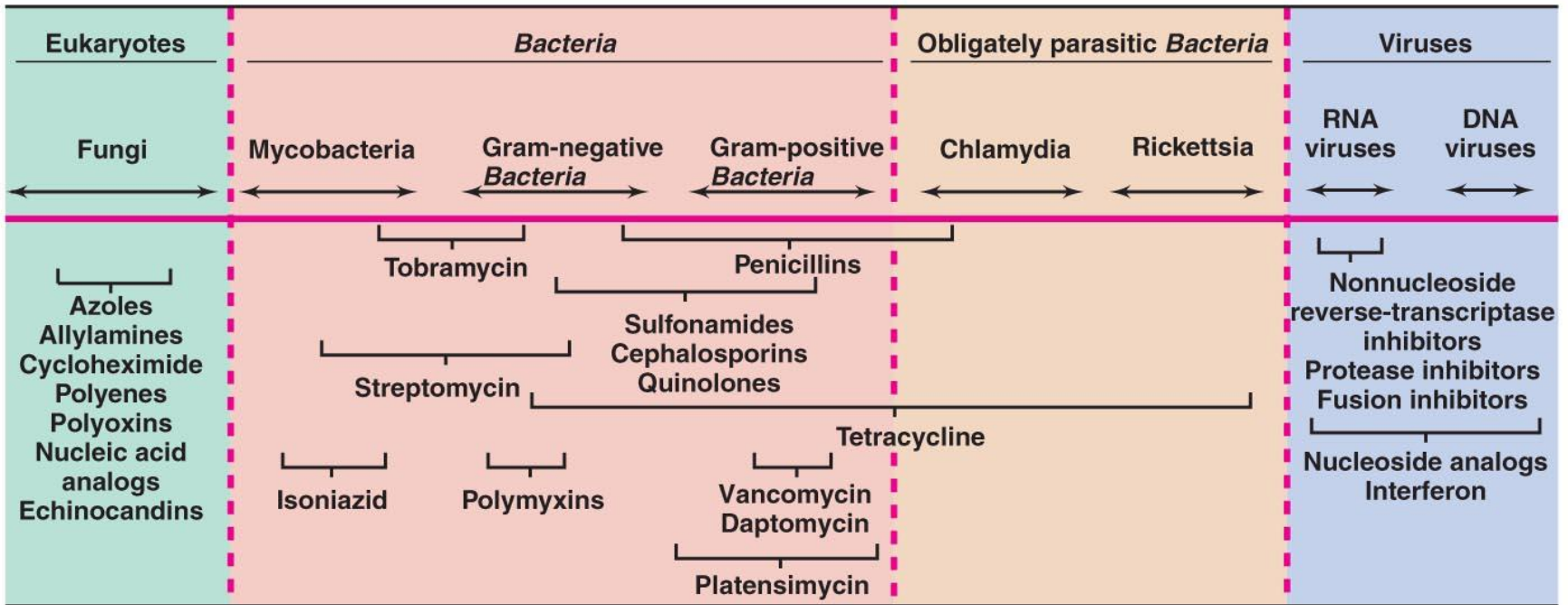


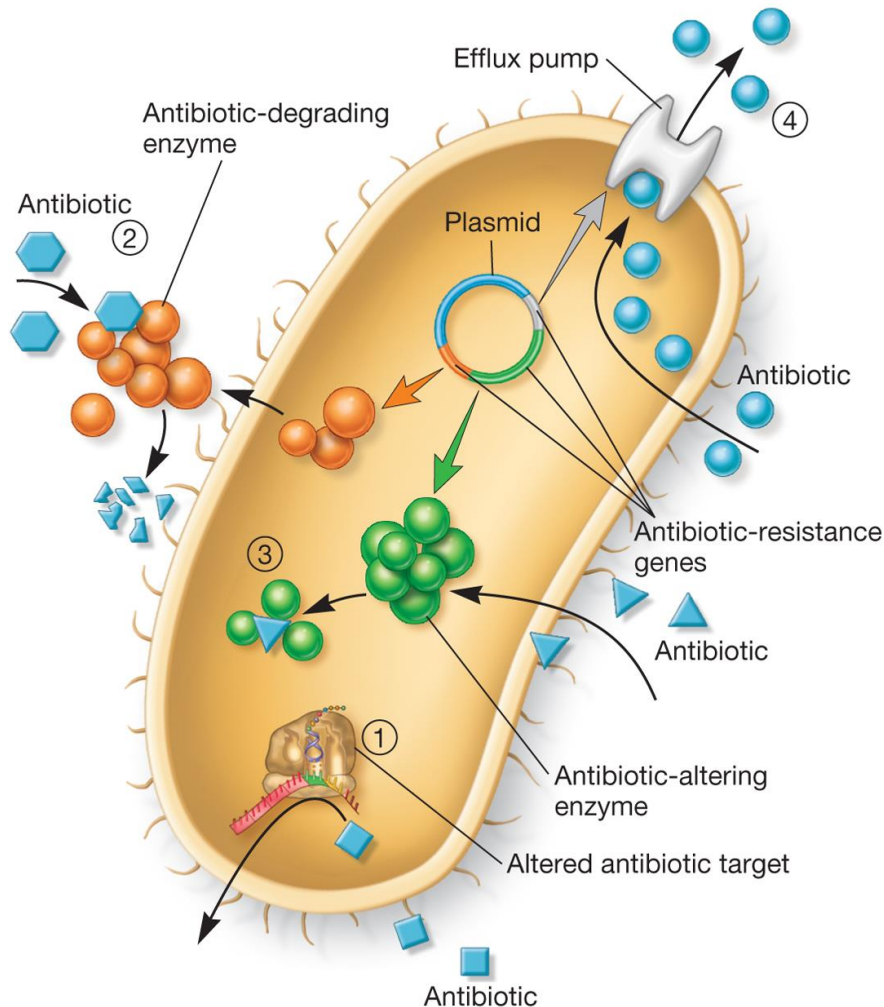
Figure 27.13

Antimicrobial Drug Resistance

- *Antimicrobial drug resistance*
 - The acquired ability of a microbe to resist the effects of a chemotherapeutic agent to which it is normally sensitive

Mechanisms of Drug Resistance

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- Prevent entrance of drug
- Drug efflux (pump drug out of cell)
- Inactivation of drug
 - chemical modification of drug by pathogen
- Modification of target enzyme or organelle
- Use of alternative pathways or increased production of target metabolite

Sites at Which Antibiotics are Attacked by Enzymes

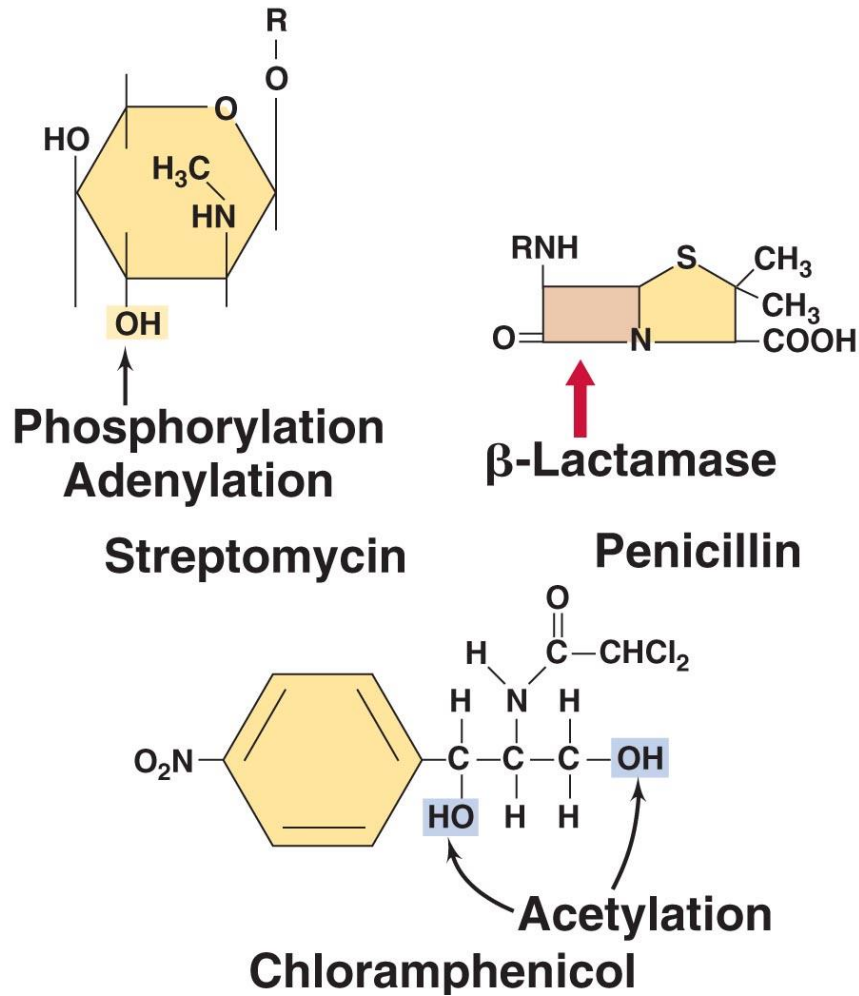


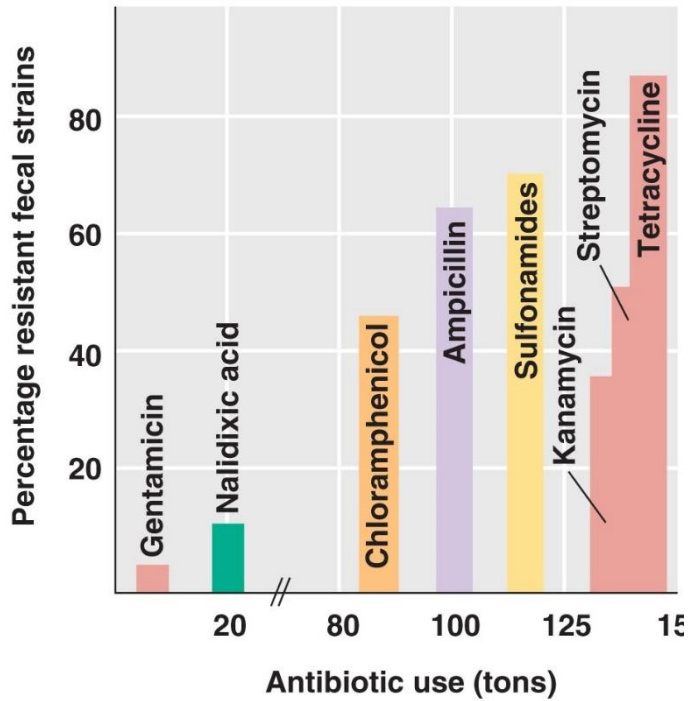
Figure 27.27

Antimicrobial Drug Resistance

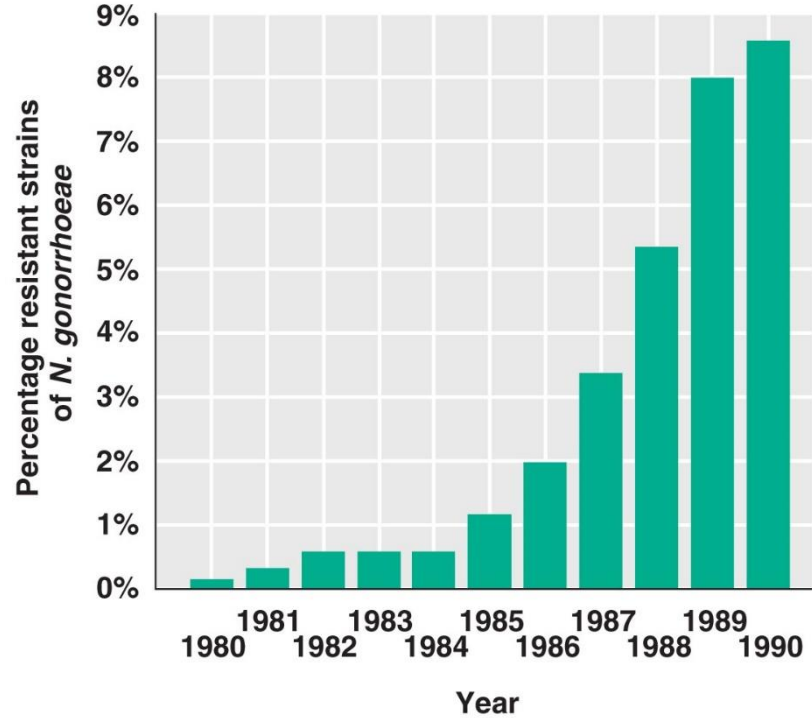
- Most drug-resistant bacteria isolated from patients contain drug-resistance genes located on **R plasmids**
- The use of antibiotics in **medicine, veterinary, and agriculture** select for the **spread of R plasmids**
 - Many examples of **overuse** of antibiotics
 - Used far more often than necessary (i.e., antibiotics used in agriculture as supplements to animal feed)



Patterns of Drug Resistance in Pathogens



(a)



(b)

Figure 27.28

Figure 27.28